



# PIN HR-3071 - TIN BRONZE - LATE BRONZE AGE - SWITZERLAND

Artefact name Pin HR-3071

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### ➤ The object



Fig 1: Pin with decorated head and round section,





Fig. 2: Green corrosion products (detail) around the head of the pin,



Fig. 3: Dense and smooth olive green corrosion products (detail) on the middle of the pin with lacunas showing the underlying metal,

2 mm

#### ▼ Description and visual observation

Description of the artefact Pin with decorated head and round section. Locally a dense and smooth olive green stratum is preserved, while green

corrosion products develop on the underlying metal (Figs. 1-3). Dimensions: L = 19.0cm; WT = 12.4g.

Type of artefact Jewellery

Origin Hauterive - Champréveyres, Neuchâtel, Neuchâtel, Switzerland

**Recovering date** Excavation 1983-1985, object from layer 1

Chronology category Late Bronze Age

chronology tpq 1050 B.C. ✓

chronology taq 800 B.C. ✓

Chronology comment

Burial conditions / environment Lake

Artefact location Laténium, Neuchâtel, Neuchâtel

Owner Laténium, Neuchâtel, Neuchâtel

Inv. number HR-3071

Recorded conservation data None.

#### Complementary information

The object was analyzed in 1987 by Schweizer. Documentation of the strata in binocular mode of the object was performed in 2022.

#### Study area(s)



Credit HE-Arc CR, N.Gutknecht/L.Rémy

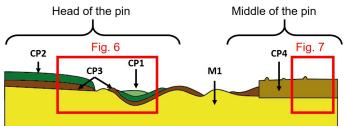
Fig. 4: Sides A and B (opposite sides) of the pin showing the XRF analysis areas (red circles),  $\,$ 

#### ★ Binocular observation and representation of the corrosion structure

The schematic representation below gives an overview of the corrosion structure encountered on the pin from a first visual macroscopic observation.

| Strata | Type of stratum   | Principal characteristics                            |  |  |  |  |  |  |  |
|--------|-------------------|--|--|--|--|--|--|--|--|
| CP1    | Corrosion product | Light green, thin, scattered, non compact, very soft |  |  |  |  |  |  |  |
| CP2    | Corrosion product | Dark green, thin, scattered, compact, very soft      |  |  |  |  |  |  |  |
| CP3    | Corrosion product | Dark brown, medium, discontinuous, compact, hard     |  |  |  |  |  |  |  |
| CP4    | Corrosion product | Olive green, medium, discontinuous, compact, hard    |  |  |  |  |  |  |  |
| M1     | Metal             | Yellow, metallic, compact, hard                      |  |  |  |  |  |  |  |

Table 1: Description of the principal characteristics of the strata as observed under binocular and described according to Bertholon's method.



Credit HE-Arc CR. N.Gutknecht.

Fig. 5: Stratigraphic representation of the corrosion structure of the pin by macroscopic and binocular observation using the Micorr application with reference to Figs. 6 and 7,

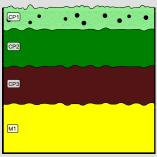


Fig. 6: Stratigraphic representation of the corrosion structure of the head of the pin (Fig. 2) observed macroscopically under binocular microscope using the MiCorr application with reference to Fig. 5. The characteristics of the strata, such as the discontinuity, are accessible by clicking on the drawing that redirects you to the search tool by stratigraphy representation, Credit HE-Arc CR, N.Gutknecht.



Fig. 7: Stratigraphic representation of the corrosion structure of the middle of the pin (Fig. 3) observed macroscopically under binocular microscope using the MiCorr application with reference to Fig. 5 where CP1 stands for CP4. The characteristics of the strata, such as the discontinuity, are accessible by clicking on the drawing that redirects you to the search tool by stratigraphy representation, Credit HE-Arc CR, N.Gutknecht.

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**Description of sample**No sample has been taken. The observation and analysis were performed directly on the object.

Alloy Tin Bronze

Technology None

Lab number of sample 85-194

Sample location None

Responsible institution None

Date and aim of sampling

Complementary information

None.

#### 

Analyses performed:

Non-invasive approach

XRF with handheld portable X-ray fluorescence spectrometer (NITON XL5). General Metal mode, acquisition time 60s (filters: Li20/Lo20/M20).

XRF analysis was carried out on two representative areas of the surface (Fig. 4). Point 1 was performed on a lacuna of the olive green corrosion layer and point 2 on the olive green corrosion layer (CP4 of Fig. 5).

The metal is presumably a tin bronze alloy with proabably some Sb and As and traces of Pb and Ag. The other elements detected are: S, Fe, Si, Zn.

Results of point 2 are very different from those of point 1, they indicate the enrichment in Fe and in S and depletion in Cu.

| Elements<br>(mass %) | Cu   |        |     | Sn     |      | s      |      | Fe     |     | Sb     |     | As     |     | Pb     |     | Ag     |     | Si     |     | Zn     | Total |
|----------------------|------|--------|-----|--------|------|--------|------|--------|-----|--------|-----|--------|-----|--------|-----|--------|-----|--------|-----|--------|-------|
|                      | %    | +/- 2σ | %   | +/- 2σ | %    | +/- 2σ | %    | +/- 2σ | %   | +/- 2σ | %   | +/- 2σ | %   | +/- 2σ | %   | +/- 2σ | %   | +/- 2σ | %   | +/- 2σ |       |
| 1                    | 87.0 | 0.2    | 8.5 | 0.05   | 1.5  | 0.04   | 0.2  | 0.01   | 0.9 | 0.02   | 0.7 | 0.03   | 0.4 | 0.02   | 0.3 | 0.01   | 0.2 | 0.06   | 0.1 | 0.03   | 99.8  |
| 2                    | 36.5 | 0.1    | 4.0 | 0.02   | 25.0 | 0.08   | 32.0 | 0.09   | 0.5 | 0.01   | 0.2 | 0.01   | 0.1 | 0.01   | 0.2 | 0.01   | 0.5 | 0.04   | 0.1 | 0.02   | 99.1  |

Table 2: Chemical composition of the surface of the pin at two representative areas shown in Fig. 5. Method of analysis: XRF.

None.

Microstructure None

First metal element Cu

Other metal elements Sn

#### Complementary information

None

CP4 (dense, smooth olive green stratum) is enriched with Fe and S and depleted in Cu. It seems to correspond to chalcopyrite (CuFeS2).

Corrosion form Multiform

Corrosion type lake patina (Schweizer 1994)

#### Complementary information

In the article "Bronze objects from Lake sites: from patina to bibliography. In: Ancient and historic metals, conservation and scientific research" (Schweizer 1994), the corrosion products of the pin 3071 (LAB MAH 85-194) were analysed with XRD. The results show that the pin contains sulfosalt (sinnerite Cu6As4S9) and copper carbonate (malachite Cu2(CO3)(OH)2) as well as copper iron sulfide (chalcopyrite CuFeS2). Sinnerite appears as dark cristals, malachite as green cristals and chalcopyrite as a brown smooth layer.

#### ♥ Synthesis of the binocular / cross-section examination of the corrosion structure

The corrosion structure has only been documented in binocular mode (Figs. 6 & 7).

### **♥** Conclusion

The pin is made from a tin bronze with possibly some Sb and As. It has been extensively documented by Schweizer to establish the lake and terrestrial patina typologies in his research paper from 1994. Chalcopyrite indicates a lake patina that was generated by the presence of sulfato-reducing bacteria

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## ▼ References

#### References on object and sample

#### Object files in MiCorr

- 1. MiCorr\_Pin or needle fragment HR-3031
- 2. MiCorr\_Tang fragment of a knife HR-6567
- 3. MiCorr\_Tang fragment of a knife HR-6246
- 4. MiCorr\_Pin HR-18152
- 5. MiCorr\_Pin HR-17773
- 6. MiCorr\_PIN HR-18603
- 7. MiCorr\_Pin HR-3389

#### References object

- 8. Rychner-Faraggi A-M. (1993) Hauterive Champréveyres 9. Métal et parure au Bronze final. Archéologie neuchâteloise, 17 (Neuchâtel), pl. 61/65.
- 9. Hochuli, S. et al. (1988) SPM III Bronzezeit , Verlag Schweizerische Gesellschaft für Ur- und Frühgschichte Basel, 76-77, 379.

#### References sample

- 10. Empa Report 137 695/1991, P.O. Boll.
- 11. Rapport d'examen, Lab. Musées d'Art et d'Histoire, Geneva GE, 87-194 à 87-197.
- 12. Schweizer, F. (1994) Bronze objects from Lake sites: from patina to bibliography. In: Ancient and historic metals, conservation and scientific research (eds. Scott, D.A., Podany, J. and Considine B.B.), The Getty Conservation Institute, 33-50.

### References on analytic methods and interpretation

13. Robbiola, L., Blengino, J-M., Fiaud, C. (1998) Morphology and mechanisms of formation of natural patinas on archaeological Cu-Sn alloys, Corrosion Science, 40, 12, 2083-2111.