

FITTING ON BACK PANEL OF A MILITARY CARRIAGE – THOMAS STEEL – MODERN TIMES – SWITZERLAND

| | |
|---------------|--|
| Artefact name | Fitting on back panel of a military carriage |
| Authors | Marianne. Senn (EMPA, Dübendorf, Zurich, Switzerland) & Christian. Degrieny (HE-Arc CR, Neuchâtel, Neuchâtel, Switzerland) |
| Url | /artefacts/371/ |

✧ The object



Credit HE-Arc CR.

Fig. 1: Steel fitting on back panel of a military carriage (©Tarchini),

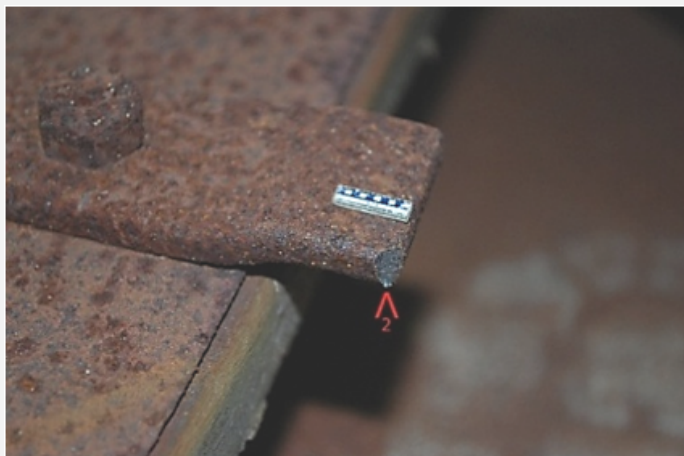
✧ Description and visual observation

| | |
|---------------------------------|---|
| Description of the artefact | Fitting on back panel, first exposed outdoors, later indoors (Fig. 1). Uniform corrosion and pitting corrosion are visible. |
| Type of artefact | Military carriage |
| Origin | Swiss Army, Thun, Bern, Switzerland |
| Recovering date | Built by Konstruktions-Werkstätte, 1918 |
| Chronology category | Modern Times |
| chronology tpq | 1918 A.D. ▼ |
| chronology taq | ----- ▼ |
| Chronology comment | 1918 |
| Burial conditions / environment | Outdoor to indoor atmosphere |
| Artefact location | Historical Swiss Army Material Foundation, Burgdorf, Bern |
| Owner | Historical Swiss Army Material Foundation, Burgdorf, Bern |
| Inv. number | n.a. |
| Recorded conservation data | Not conserved |

Complementary information

Nothing to report.

Study area(s)



Credit HE-Arc CR.

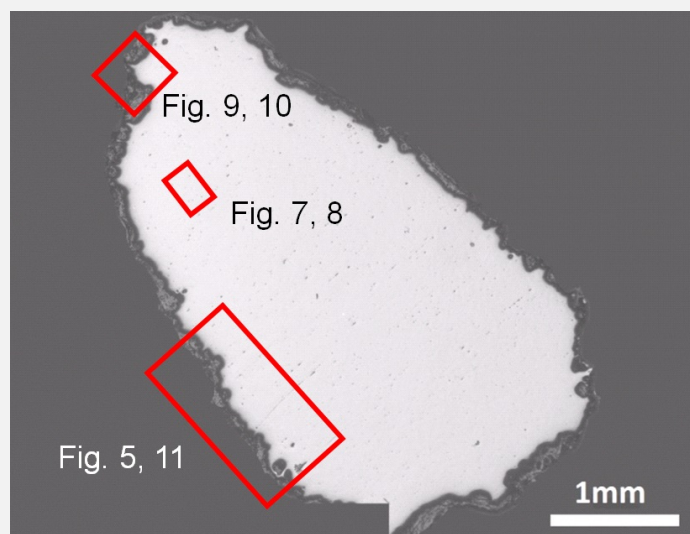
Fig. 2: Location of sampling area,

Binocular observation and representation of the corrosion structure

Stratigraphic representation: none.

MiCorr stratigraphy(ies) – Bi

Sample(s)



Credit HE-Arc CR.

Fig. 3: Micrograph of the cross-section showing the location of Figs. 5 and 7 to 11,

| | |
|--------------------------|--|
| Description of sample | This sample is a cut from the corner of one of the two fittings on the back panel (Fig. 2). The metal is covered by a thin corrosion layer (Fig. 3). |
| Alloy | Thomas steel |
| Technology | Piled from several strips, hot rolled and annealed |
| Lab number of sample | POINT-Fe2 |
| Sample location | Empa (Marianne Senn) |
| Responsible institution | Historical Swiss Army Material Foundation, Burgdorf, Bern |
| Date and aim of sampling | 05/2009 metallography |

| |
|---------------------------|
| Complementary information |
| Nothing to report. |
| ✧ Analyses and results |

| |
|---|
| Analyses performed: |
| Metallography (nital etched after etching with Oberhoffer’s reagent), Vickers hardness testing, LA-ICP-MS, SEM/EDS. |

| |
|-------------------------|
| ✧ Non invasive analysis |
| |

| |
|---------|
| ✧ Metal |
| |

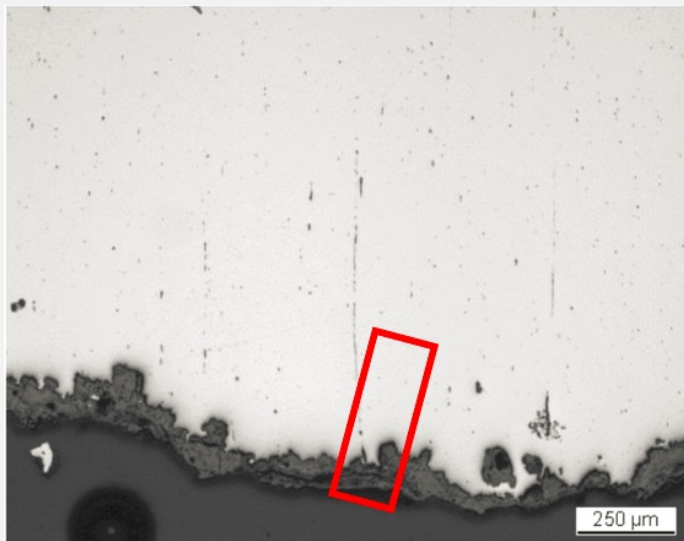
The remaining metal is a Mn-rich soft steel (C content around 0.1 mass%) containing manganese sulphide inclusions with a varying Fe content (Tables 1 and 2). The numerous inclusions form parallel rows (Fig. 5). This orientation is typical for hot rolled metal. After etching with Oberhoffer’s reagent, three main welding seams (P-rich) become visible (Fig. 6). Near the surface they are also outlined by corrosion (Figs. 5 and 11). After nital etching, the metal shows a ferritic structure with tertiary cementite and lamellar pearlite at the grain boundaries (Figs. 7 and 8). The grains are small with an ASTM grain size of 10 and are recrystallized due to annealing after hot rolling. The average hardness of the metal is HV1 165. The hardness is slightly high for such a structure and this is due to the Mn content of the metal. The chemical composition, especially the Mn content and the presence of carbo-nitrides (not analysed here), is typical for Thomas steel.

| Elements | Ni/Co | Al | P | Ti | V | Cr | Mn | Co | Ni | Cu | As | Mo | Ag | Sn | Sb | W | C* mass% |
|-------------------------|-------|----|-----|----|---|-----|------|-----|-----|-----|-----|----|----|----|----|---|----------|
| Median (mg/kg) | 2.4 | < | 300 | < | < | 140 | 3600 | 200 | 480 | 140 | 700 | 10 | < | 10 | 10 | < | <0.1 |
| Detection Limit (mg/kg) | | 5 | 82 | 10 | 2 | 13 | 2 | 1 | 3 | 1 | 3 | 3 | 1 | 1 | 1 | 4 | |
| RSD % | 1 | - | 8 | - | - | 3 | 7 | 2 | 1 | 6 | 3 | 7 | - | 7 | 8 | - | |

*visually estimated
 Table 1: Chemical composition of the metal. Method of analysis: LA-ICP-MS. Lab Inorganic Chemistry, ETH.

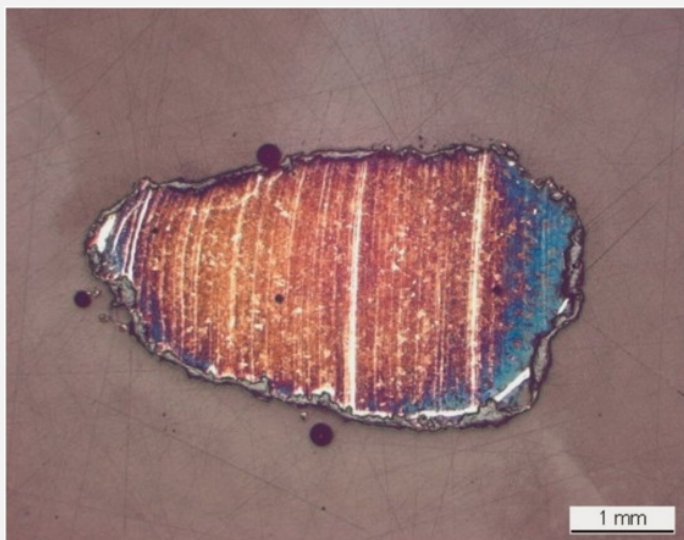
| Elements | S | Mn | Fe | Cu | Total |
|------------|----|----|----|-----|-------|
| Inclusions | 26 | 46 | 31 | 2.2 | 105 |

Table 2: Chemical composition (mass %) of the inclusions. Method of analysis: SEM/EDS, Laboratory of Analytical Chemistry, Empa.



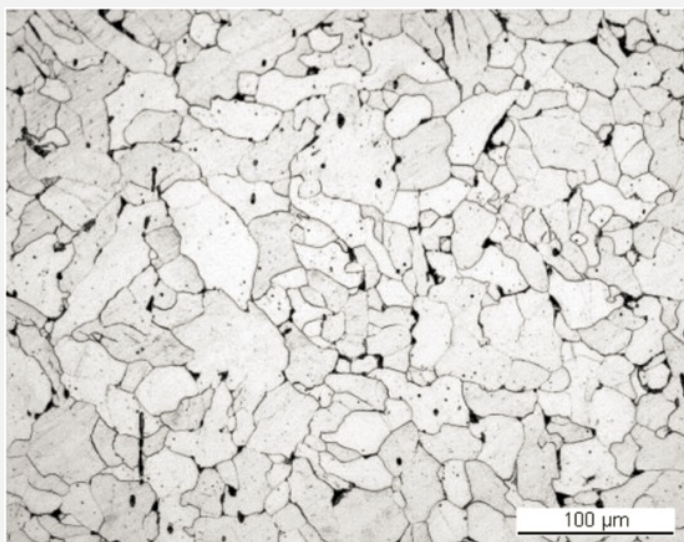
Credit HE-Arc CR.

Fig. 5: Micrograph of the metal sample from Fig. 3 (rotated by 270°, detail), unetched, bright field. We observe the metal including welding seams outlined by corrosion products and MnS inclusions. The mapped area of Fig. 11 is marked by the rectangle,



Credit HE-Arc CR.

Fig. 6: Micrograph of the metal sample from Fig. 3, etched with Oberhoffer's reagent, bright field. We observe three P-rich welding seams in white,



Credit HE-Arc CR.

Fig. 7: Micrograph of the metal sample, nital etched, bright field. We observe tertiary cementite (black) on the grain boundaries,

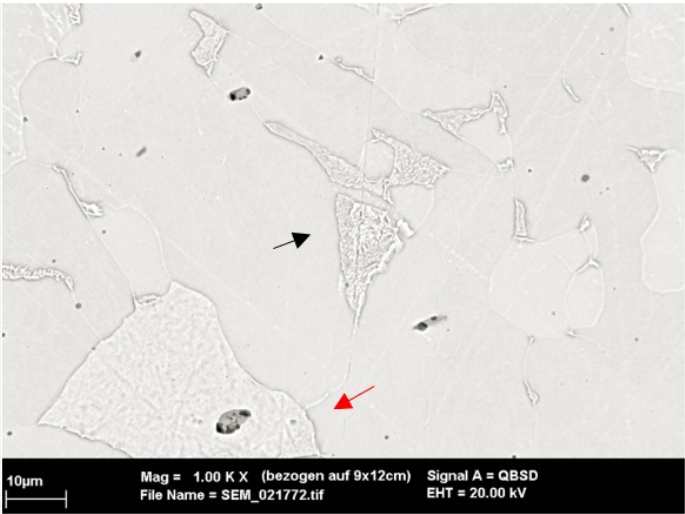


Fig. 8: SEM image of the metal sample from Fig. 3 (detail), BSE-mode, nital etched, bright field. The white ferrite grains contain lamellar pearlite (black arrow) and lenticular grey MnS inclusions (red arrow) in between the grain boundaries,

| | |
|----------------------|--|
| Microstructure | Recrystallized grain structure with tertiary cementite |
| First metal element | Fe |
| Other metal elements | C, Mn |

Complementary information

Nothing to report.

Corrosion layers

The average thickness of the corrosion products is about 80µm (Figs. 5 and 9). In bright field they appear grey, marbled and heavily cracked (Fig. 9). Under polarised light, the corrosion products appear orange to dark-brown (Fig. 10). At the metal - corrosion products interface they are dark-brown (CP3). The middle part (CP2) is red-orange and the outer part is bright orange (CP1). The elemental mapping of the corrosion layers shows no distinctive stratification, but areas near the metal - corrosion crust interface (CP3) as well as the top surface of the corrosion layer (CP1) seem to have a lower O content (Fig. 11). The O content indicates the presence of iron hydroxides (Table 3). Soil materials (such as rock fragments and dust) are found in the welding seams near the surface.

| Elements Location | O | S | Mn | Fe | Total |
|-----------------------------|----|-----|-----|----|-------|
| In welding seam | 34 | < | < | 67 | 102 |
| Inner corrosion layer (CP3) | 38 | 0.7 | 0.8 | 66 | 106 |

Table 3: Chemical composition (mass %) of the corrosion layer (from Fig. 11). Method of analysis: SEM/EDS, Laboratory of Analytical Chemistry, Empa.

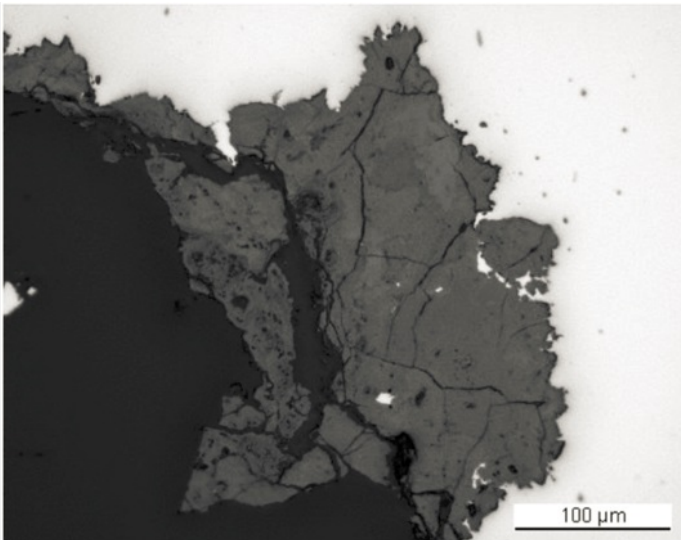


Fig. 9: Micrograph showing the metal - corrosion crust interface from Fig. 3 (rotated by 270°, detail), unetched, bright field. We observe in white the metal, in grey the corrosion products and in black the resin,

Credit HE-Arc CR.

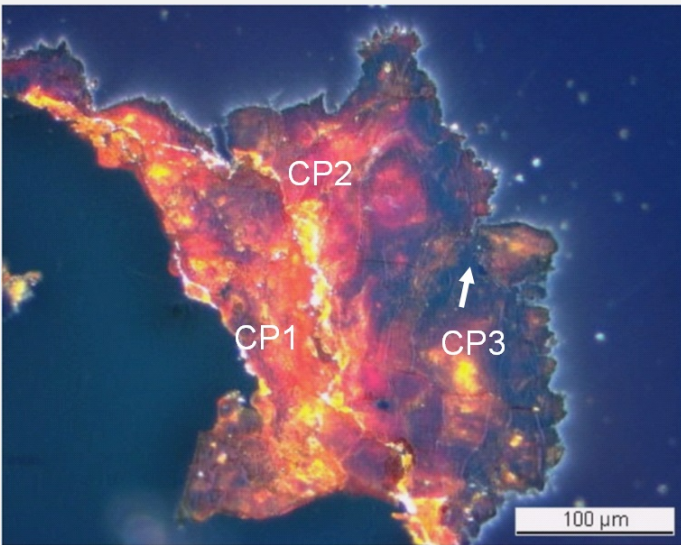


Fig. 10: Micrograph (same as Fig. 9) and corresponding to the stratigraphy of Fig. 4, unetched, polarised light. The corrosion products are dark-brown at the metal - corrosion crust interface and red-orange on the outside,

Credit HE-Arc CR.

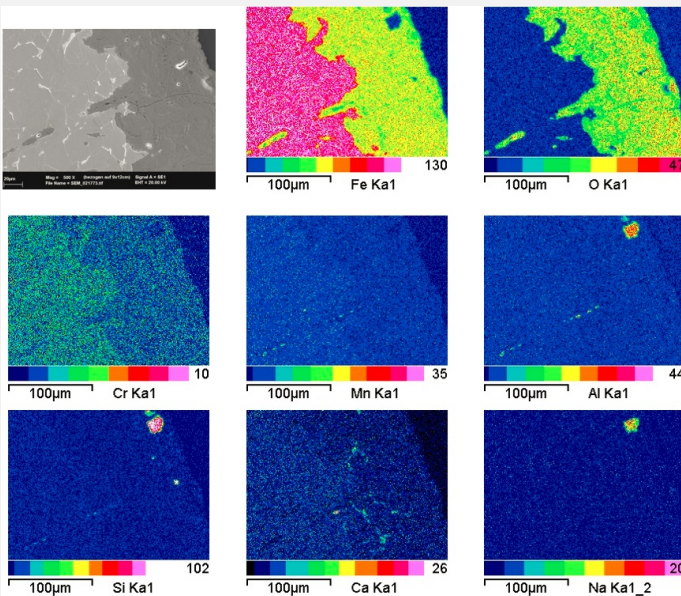


Fig. 11: SEM image, BSE-mode, and elemental chemical distribution of the selected area from Fig. 5 (rotated by 270°, detail). Method of examination: SEM/EDS, Laboratory of Analytical Chemistry, Empa.

Credit HE-Arc CR.

Corrosion form Uniform - transgranular

Corrosion type ?

Complementary information

Nothing to report.

✧ MiCorr stratigraphy(ies) – CS

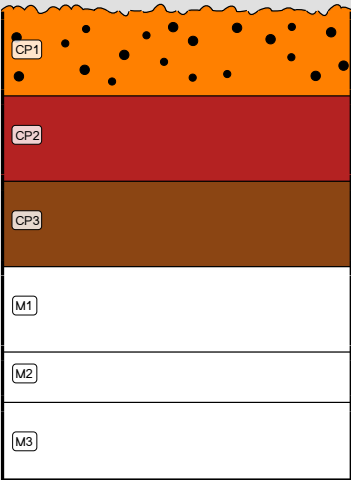


Fig. 4: Stratigraphic representation of the object in cross-section using the MiCorr application. This representation showing for the metal part a welding seam (M2) can be compared to Fig. 10, Credit HE-Arc CR.

✧ Synthesis of the binocular / cross-section examination of the corrosion structure

Corrected stratigraphic representation: none.

✧ Conclusion

The fitting was produced from Mn-containing Thomas steel. It was forged out of four strips, hot rolled and annealed. The corrosion contains only few external markers such as sand grains and dust particles in the outermost layers. The presence of soil materials in the welding seams near the surface could be due either to the corrosion progress (by diffusion through the corrosion crust) or to the manufacturing process.

✧ References

References on object and sample

References object

1. Degriigny, C. (2011) Protection temporaire d'Objets métalliques base fer et cuivre à l'aide d'Inhibiteurs de corrosion Non Toxiques : application aux objets patrimoniaux techniques et scientifiques de grandes dimensions exposés en atmosphère non contrôlée, rapport interne HE Arc CR.

References sample

2. Rumo, L. (2009) Rapport Empa.

References on analytic methods and interpretation

ASTM E112-13: Standard Test Methods for Determining Average Grain Size.