



CIRCULAR DOOR KNOCKER STRIKING PLATE - CU ALLOY - MODERN TIMES - SWITZERLAND

Artefact name Circular door knocker striking plate

Authors Marianne. Senn (EMPA, Dübendorf, Zurich, Switzerland) & Christian. Degrigny (HE-Arc CR, Neuchâtel, Neuchâtel,

Switzerland)

Url /artefacts/437/

▼ The object

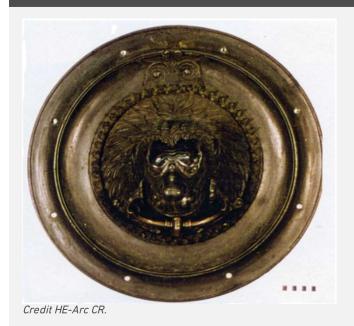


Fig. 1: Copper alloy circular base from a door knocker (after Benoît, 1994),

▼ Description and visual observation

Description of the artefact Circular base no 2 (from a set of two) of a door knocker with a protruding lion head holding a hinge in his

fangs, Ø.ext = 64cm. The circular base is covered by a gilded (?) decorative relief (Fig. 1, Benoit, 1994, 6).

Type of artefact Architectural element

Origin Door knocker of Lausanne cathedral, Lausanne, Vaud, Switzerland

Recovering date None

Chronology category Modern Times

chronology tpq 1601 A.D. ▼

chronology taq 1700 A.D. ✓

Chronology comment 17th Century AD

Burial conditions / environment Outdoor atmosphere

Artefact location Lausanne cathedral, Lausanne, Vaud

Owner Lausanne cathedral, Lausanne, Vaud

Inv. number None

Complementary information

Nothing to report.

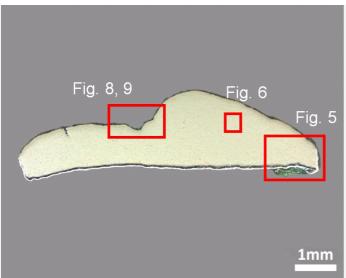


Fig. 2: Possible location of sampling area,

Stratigraphic representation: none.

\forall Sample(s)

Fig. 3: Micrograph of the cross-section showing the location of Figs. 5 to 6 and 8 to 9, $\,$



Credit HE-Arc CR.

Description of sampleNo information is given on where the sample has been taken. A supposed location is however illustrated in

Fig. 2. The polished sample shows a well preserved metal surface with some cracks covered by a thin

corrosion crust (Fig. 3).

Alloy Cu Alloy

Technology Rolled, annealed after cold working

Lab number of sample MAH 94-156-001

Sample location Musées d'art et d'histoire, Genève, Geneva

Responsible institution Musées d'art et d'histoire, Genève, Geneva

Date and aim of sampling 1994, metallography and presence or not of a gilding treatment

Complementary information

Nothing to report.

Analyses performed:

Metallography (etched with ferric chloride reagent), Vickers hardness testing, LA-ICP-MS, SEM/EDS, Raman spectroscopy.

▼ Non invasive analysis

× Meta

Analysis by LA-ICP-MS (Table 1) indicates that the metal is a copper alloy. The high relative standard deviations (RSD) result from the heterogeneity of the metal (inclusions). The alloy differs from the quaternary alloy given in the report for the circular base considered here (Benoît 1994, p.18). It contains numerous inclusions of copper oxides (Figs. 5, 6, 8 and 9, Table 2) and heavy metals (Sn, Pb, Fig. 9 and Table 2) that are residues from the manufacturing process. In polarized light, the copper oxide inclusions appear red (Figs. 5 and 8) while with SEM in BSE-mode they look light-grey (Fig. 9). Heavy metals inclusions appear white with SEM (BSE-

mode, Fig. 9). Due to the rolling process the copper oxide inclusions are parallel to the longitudinal orientation of the sample (Figs. 5 and 6). They were identified as cuprite by Raman spectroscopy (Fig. 7). The etched alloy shows a structure of polygonal grains with annealing twins (Fig. 6). The grain size is variable. The average hardness of the alloy is HV1 80, which is quite high compared to the average hardness of a pure annealed copper of around HV1 40-50 (Schumann 1991, 627).

Elements	Cu	Sb	Pb	Ag		Sn	Zn		As	
mass%	99.6	0.2	0.12	0.1	0.004	<	<	<	<	<
RSD %	0.3	112	132	1	112					

Table 1: Chemical composition of the metal. Method of analysis: LA-ICP-MS, Laboratory of Basic Aspects of Analytical Chemistry at the Faculty of Chemistry, University of Warsaw, PL.

Elements	0	Cu	Fe		Sn	Sb	Pb	Total
Light-grey inclusion	10	86	<	<	<	<	<	96
White inclusion 1	18	48	17	3	11	0.9	0.8	105
White inclusion 2	13	8	<	<	<	34	44	99

Table 2: Chemical composition (mass %) of the inclusions (light-grey and white in Fig. 9) in the metal. Method of analysis: SEM/EDS, Laboratory of Analytical Chemistry, Empa.

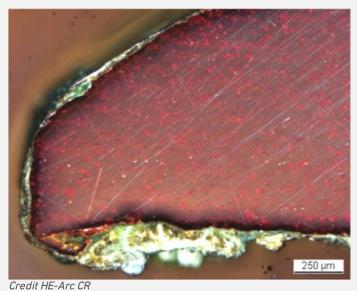


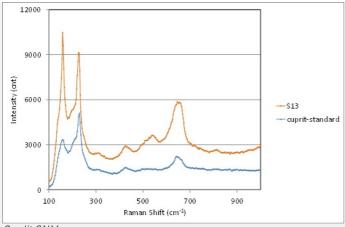
Fig. 5: Micrograph of the metal sample from Fig. 3 (reversed picture), unetched, polarised light. The red copper oxide inclusions form rows in the copper alloy matrix.,

All and

Tredit HE-Arc CR.

Fig. 6: Micrograph of the metal sample from Fig. 3 (detail), etched, bright field. The metal shows a structure of polygonal and twinned grains. Copper oxide inclusions are visible as dark spots,

Fig. 7: Raman spectrum of a red inclusion (S13) of Fig. 5 compared to a cuprite standard spectrum. Settings: laser wavelength 532nm,



acquisition time 100s, one accumulation, filter D2 (0.75-0.8mW), hole 500, slit 80, grating 600. Method of analysis: Raman spectroscopy. Lab of Swiss National Museum, Affoltern a. Albis ZH,

Credit SNM.

Microstructure Polygonal and twinned grains, elongated inclusions

First metal element Cu

Other metal elements Ag, Sb, Pb

Complementary information

Nothing to report.

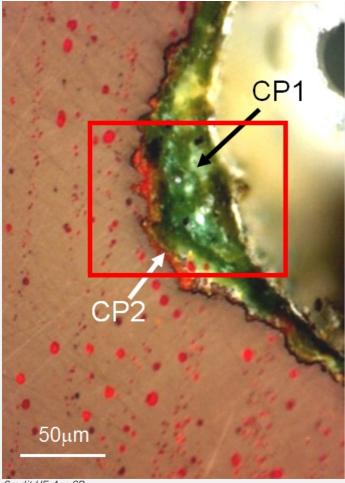
▼ Corrosion lavers

The average thickness of the corrosion crust is about $90\mu m$, but may be thinner or thicker depending on the area of origin (Fig. 5). Under polarized light, we observe a multilayer system with a dense and thin inner red corrosion layer (CP2, probably cuprite/Cu20) on the metal (Fig. 8) containing chlorides (Fig. 9 and Table 3). This is followed by bands of heterogeneous green corrosion products (CP1, possibly copper chloride / sulphate / carbonate). This outer corrosion layer is contaminated with atmospheric components (Si, Al, Cl, C and O and perhaps gypsum particles (CaSO4)) (Fig. 9).

Elements	0	Si	Cl	Cu	Total
CP2, inner red layer	11	<	23	60	94

Table 3: Chemical composition (mass %) of the inner corrosion layer (CP2). Method of analysis: SEM/EDS, Laboratory of Analytical Chemistry, Empa.

Fig. 8: Micrograph of the metal sample from Fig. 3 (rotated by 90°) and corresponding to the stratigraphy of Fig. 4, polarised light. Metal with red copper oxide inclusions. The black particles in the green corrosion layer are quartz inclusions. The mapped area of Fig. 9 is marked by a rectangle,



Credit HE-Arc CR.

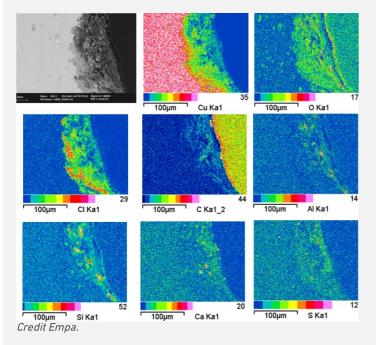


Fig. 9: SEM image, BSE-mode, and elemental chemical distribution of the selected area of Fig. 8. Method of examination: SEM/EDS, Laboratory of Analytical Chemistry, Empa,

Corrosion form

Uniform - pitting

Corrosion type

Mostly type I with locally type II (Robbiola)

Complementary information

Nothing to report.

(PI) (P2) (M1)	Fig. 4: Stratigraphic representation of the object in cross-section using the MiCorr application. The characteristics of the strata are only accessible by clicking on the drawing that redirects you to the search tool by stratigraphy representation. This representation can be compared to Fig. 8, Credit HE-Arc CR.
▼ Synthesis of the binocular / cross-section examination	n of the corrosion structure

♥ Conclusion

Corrected stratigraphic representation: none.

The circular base of the door knocker consists of a rolled copper alloy that was annealed after cold working. The metal is rich in copper oxide inclusions. The corrosion is most likely composed of cuprite in proximity to the metal surface, followed by green copper corrosion products (possibly chlorides, carbonates or sulphates) on the outer surface which is also contaminated with gypsum and airborne particles. The presence of chlorides in the inner corrosion layers could be explained by contamination through handling. Soil elements are probably originating from airborne dust, whereas sulphur could come from urban SO2 pollution. No trace of gilding was observed. In this case the corrosion is mostly of type 1, but can be locally of type 2 after Robbiola et al. 1998.

▼ References

References on object and sample

1. Benoît, C. (1994) Cathédrale de Lausanne: conservation de deux appliques en bronze à tête de lion avec anneau mobile et encadrement circulaire. Rapport de travail, non publié.

2. Rapport d'examen, Laboratoire Musées d'Art et d'Histoire, Genève No 94-156-1/2 (1998).

References on analytic methods and interpretation

3. Robbiola, L., Blengino, J-M., Fiaud, C. (1998) Morphology and mechanisms of formation of natural patinas on archaeological Cu-Sn alloys, Corrosion Science, 40, 12, 2083-2111.

4. Schumann, H. (1991) Metallographie, Leipzig.