



BUGATTI COOLING SYSTEM PIPE - AL ALLOY - FRANCE

Artefact name Bugatti cooling system pipe

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▼ The object



Fig. 1: Bugatti water pipe, unknown model,

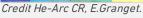




Fig. 2: Examples of water pipes located on a Type 37 engine,

▼ Description and visual observation

Description of the artefact This section of a pipe was given to the MNAM as study material, without indication of provenance. The

A.D. V

pipe is angled, sharply cut on one end and heavily corroded on the other, the inside being clugged by a

brittle yellowish deposit.

Looking at its shape, it is safe to assume that this is an exit pipe from the upper part of the engine (Fig. 2

and complementary information below).

Type of artefact Technical object

Origin MNAM (Musée National de l'Automobile de Mulhouse), Mulhouse, Alsace, France

Recovering date None **Chronology category** None

chronology tpq 1920

chronology taq 1930 A.D. ∨

Chronology comment

Burial conditions / environment

Outdoor to indoor atmosphere

Artefact location MNAM (Musée National de l'Automobile de Mulhouse), Mulhouse, Alsace

Owner MNAM (Musée National de l'Automobile de Mulhouse), Mulhouse, Alsace

Inv. number None

Recorded conservation data None

Complementary information

A combustion engine transforms thermal energy into kinetic energy. This process is imperfect and releases a lot of heat through the block of the engine. Therefore, these parts need to be cooled down. A cooling system circulating water between the block [hot] and a heat exchanger (or radiator) [cold] is frequently used to fulfill this function (Poulain, 1995, p.86). Most of the time, the Bugatti inline cylinder engines have aluminium water pipes entering the block from below and exiting it from above, as shown on Fig.2.

This artefact is part of the "Materials for study Library" that the museum collected. It can therefore be sampled.

Fig. 3: Heavily corroded side of the Bugatti water pipe showing the sampling area indicated in red,







Credit He-Arc CR, E.Granget.

The schematic representation below gives an overview of the corrosion layers encountered on the Bugatti pipe.

The stratum M is an aluminium alloy. Three CP strata have been identified. CP3 has a greyish mate color and is covering all of the corroded side of the pipe. Scattered over this uniform layer, blue-green (CP2) and white (CP1) spots of corrosion products can be found.

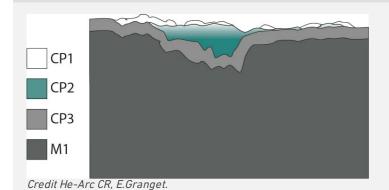


Fig. 4: Stratigraphic representation of the corrosion structure of the external surface of the water pipe in cross section,

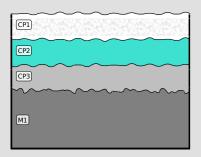


Fig. 5: Stratigraphic representation of the Bugatti water pipe under binocular using the MiCorr application. The characteristics of the strata are only accessible by clicking on the drawing that redirects you to the search tool by stratigraphy representation. This representation can be compared to Fig. 4, Credit HE-Arc CR, E.Granget.

Fig. 6: Micrograph of the sample taken from the corroded exit pipe of the Bugatti water pump (Fig. 3), showing the location of Figs. 8-9







Credit He-Arc CR, E.Granget.

Description of sample A radial section of the pipe has been sampled and embedded as shown on Fig.6.

> The cross-section shows not only the metal sampled but the accompanying thick yellow deposit that was clugging the pipe. Pitting corrosion develops all along the profile of the sample. A selected area (in

red) is further investigated in Figs. 8-9 and 12-15.

Alloy Al Alloy

Technology Cast

Lab number of sample

Sample location HE-Arc CR, Neuchâtel, Neuchâtel

Responsible institution MNAM (Musée National de l'Automobile de Mulhouse), Mulhouse, Alsace

Date and aim of sampling 31.12.2019 - Sampled for Metallography

Complementary information

The yellow deposit that was clugging the pipe couldn't be properly analysed (see Analyses and results below). Indeed, the deposit was way softer than the metal and suspected to contain organic compounds coming from the engine's coolant.

Analyses performed on the pipe

XRF with portable X-ray fluorescence spectrometer (Niton XL3t 950 Air Goldd+ analyser Thermo Fischer (voltage 50V, General metals mode with acquisition times 20s (Main) /20s (Low) /20s (Light).

Sample preparation

Due to a significant difference between the metal and the deposit hardness, and in order to preserve the corrosion layers, the surface preparation was realized with an alcoholic lubricant instead of water. Thus, the quality for the metal and corrosion layers have been priviledged over the deposit, as the alcohol might have corrupted the suspected organic compounds of the deposit. (See Sample: Complementary information above).

Analyses performed on the cross-section sampled from the pipe

Metallography (unetched), BF and DF imaging.

SEM-EDS (20kV), SE and BSE imaging and semi-quantitative EDX analysis.

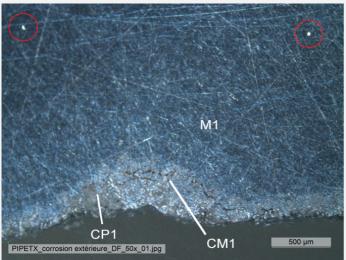


The metal is slighly porous (Fig. 8, red circles). Its dendritic microstructure is revealed by observation in bright field (Fig. 9) and with SEM (Fig. 10).

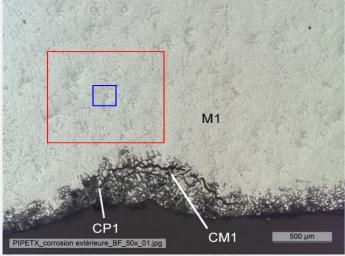
Element	mass %
Al	89
Cu	7
Zn	2.5
Fe	1
Si	0.5

Table 1: Chemical composition (mass %) of the metal the water pump pipe. Method of analysis: SEM/EDS, HE-Arc Ingénierie, S.Ramseyer.

Point analyses of the main dendritic phase showed that it is made of Al, Cu and Zn (Table 1 and Fig. 11). Elemental chemical distribution with EDS allowed to identified the following repartition (Fig. 12): Zn is uniformely present in the alloy. There are two interdendritic phases (Fig. 13): a eutectic phase containing Fe and Cu, and a second phase containing mostly Cu. Additionally Sn and Si precipitates could be identified.



Credit HEI Arc, S.Ramseyer / Edit: He-Arc CR, E.Granget.



Credit HEI Arc, S.Ramseyer / Edit: He-Arc CR, E.Granget.

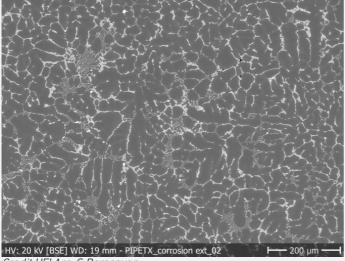
Fig. 8: Micrograph of the selected area shown on Fig.6 (rotated by 90°), unetched, dark field with indication of pores (red circles),

Fig. 9: Micrograph similar to Fig. 8, unetched, bright field showing the locations of Fig. 10 (in red) and Fig. 11 (in blue),

Fig. 10: SEM image (BSE mode) of the metal sample from Fig. 9 showing the microstructure of the pipe. Dark grey = main phase (Al), medium grey = interdendritic phase 1 (eutectic Fe, Cu), light grey = interdendritic phase 2 (Cu) and white = precipitates (Si, Sn),







Credit HEI Arc, S.Ramseyer.

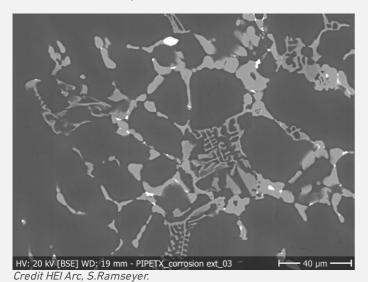


Fig. 11: Close-up image of Fig. 10,

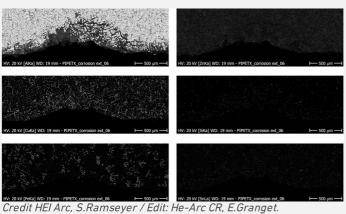
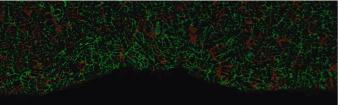


Fig. 12: SEM image and EDS elemental chemical distribution of main and interdendritic phases of the pipe. 20kV,



HV: 20kV [R:FeKa; V:CuKa; B:SnLa] - PIPETX_(← 500 µm → Credit HEI Arc, S.Ramseyer.

Fig. 13: Fe (red), Cu (green) and Sn (blue) chemical distribution of the area of Fig. 9. Method of analysis: SEM-EDS. Lab. of Electronic Microscopy and Microanalysis, Néode, HEI Arc,

Microstructure Dendritic structure

First metal element





The BF and DF images (Figs. 8 and 9) show a few big cracks at the interface between CM1 and M1. The SEM image shows that they expand in a network of additional microcracks (Fig. 14). The pipe suffers from a uniform interdendritic corrosion. The Al, Cu and Zn phase is oxidising preferentially, developing then aluminium oxides, versus the interdendritic phases. In some places, pits of corrosion formed. SEM cartography showed a concentration of Cl at the base of the pit (Fig. 14: Green).

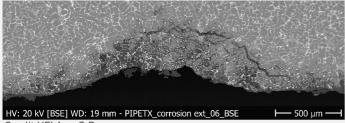
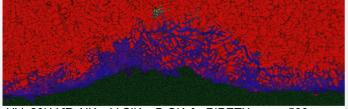


Fig. 14: SEM image of the pitting corrosion of the pipe. BSE, 20kV. Black = Cracks, Dark grey = oxydation, Medium grey = main phase, Light grey = interdendritic phase,

Credit HEI Arc, S.Ramseyer.



HV: 20kV [R:AlKa; V:ClKa; B:OKa] - PIPETX_cor ← 500 µm → Credit HEI Arc, S.Ramseyer.

Fig. 15: SEM cartography of a pit of corrosion developing on the pipe. 20kV. Red = Al, Green = Cl, Blue = O,

Corrosion form Multiform - pitting

Corrosion type None

★ MiCorr stratigraphy(ies) - CS

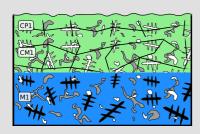


Fig. 7: Stratigraphic representation of the sample taken from the Bugatti water pipe in cross-section (dark field) using the MiCorr application. The characteristics of the strata are only accessible by clicking on the drawing that redirects you to the search tool by stratigraphy representation. This representation can be compared to Fig. 8 and 9, Credit HE-Arc CR, E.Granget.

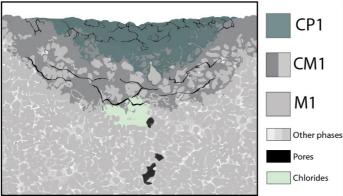
♥ Synthesis of the binocular / cross-section examination of the corrosion structure

The schematic representation of corrosion layers of Fig. 4 integrating additional information based on the analyses carried out is given in Fig. 16.

Fig. 16: Improved stratigraphic representation of the exit pipe,







Credit He-Arc CR, E.Granget.

∀ Conclusion

This angled water pipe, coming from a Bugatti engine's cooling system, is made of cast Al, Cu, Zn, Fe alloy and shows a slightly porous dendritic microstructure. There are two interdendritic phases, a eutectic Fe, Cu phase and a Cu phase, as well as precipitates of Si and Sn.

One end of the pipe is heavily corroded and entirely clugged by corrosion products and other deposits on the inside. This deposit couldn't be analyzed.

The external surface shows a uniform interdendritic corrosion preferentially consuming the dentritic phase (Al, Cu, Zn). Pitting corrosion has been identified, with Cl pockets at the base of the pit.

▼ References

References on object and sample

References object

- 1. Poulain, P. and J-M. (1995) Voitures de collection : Restauration Mécanique Editions Techniques pour l'Automobile et l'Industrie (ETAI), Paris.
- 2. Granget, E. (2020).La corrosion des alliages d'aluminium des circuits de refroidissement à eau de véhicules en contexte patrimonial : Utilisation d'outils open-access dans l'établissement d'un diagnostic des altérations d'un corpus de véhicules conservés au Musée National de l'Automobile de Mulhouse (Collection Schlumpf), Rapport interne MNAM

References sample

3. Granget, E. (2020).La corrosion des alliages d'aluminium des circuits de refroidissement à eau de véhicules en contexte patrimonial : Utilisation d'outils open-access dans l'établissement d'un diagnostic des altérations d'un corpus de véhicules conservés au Musée National de l'Automobile de Mulhouse (Collection Schlumpf), Rapport interne MNAM

References on analytic methods and interpretation

- 4. Vargel, C. (2004) Corrosion of Aluminium, Elsevier.
- 5. Degrigny, C. (2018) Etude, identification des objets en aluminium patrimoniaux et classification de leurs formes de corrosion - projet EtICAL, rapport interne HE-Arc CR.



