CHISEL POINT DEV 995 1069 PR – EUTECTOID TO HYPEREUTECTOID STEEL – EARLY MEDIEVAL TIMES – SWITZERLAND

Artefact name Chisel point DEV 995 1069 PR

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Url /artefacts/700/

The object



Credit HE-Arc CR.

Fig. 1: Chisel point (after Eschenlohr et al. 2007, 275, 314-315),

Description and visual observation

Description of the artefact The rectangular section of a massive moil chisel fragment with an incomplete tip and missing shaft (Fig. 1). Dimensions: L = 3.7cm; WT = 16g.

Type of artefact Tool

Origin Settlement Develier, Courtételle, Jura, Switzerland

Recovering date Excavated in 1995, farm 2 and workshop area 1

Chronology category Early medieval times

chronology tpq 550 A.D.

chronology taq 650 A.D.

Chronology comment

Burial conditions / environment Soil

Artefact location Office de la Culture, Porrentruy, Jura

Owner Office de la Culture, Porrentruy, Jura

Inv. number DEV 995/1069 PR

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Recorded conservation data	Conserved between 1995 and 2000: desiccation at 80°C, mechanical cleaning, passivation with tannic acid and surface protection with Paraloid B72® (Eschenlohr et al. 2007, 75).

Complementary information

Nothing to report.

Study area(s)



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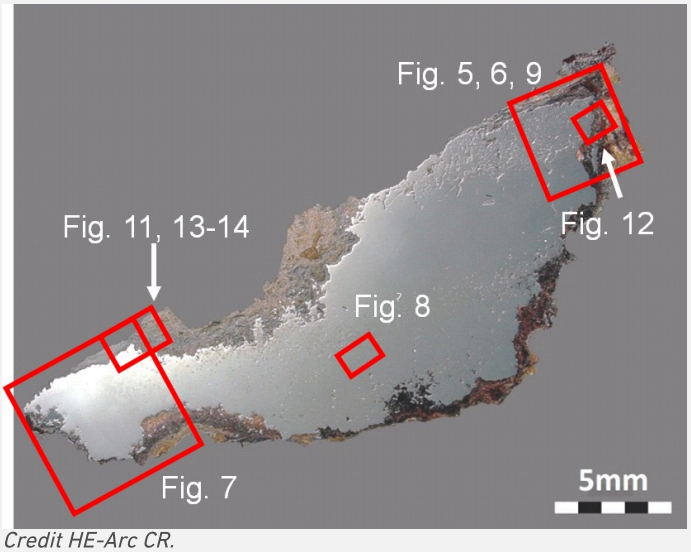
Fig. 2: Location of sampling area,

Binocular observation and representation of the corrosion structure

Stratigraphic representation: none.

MiCorr stratigraphy(ies) – Bi

Sample(s)



Credit HE-Arc CR.

Fig. 3: Micrograph of the cross-section of the chisel point showing the location of Figs. 5 to 9 and 11 to 14,

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Description of sample	A longitudinal cut from a chisel point (Figs. 2 and 3) undertaken after restoration. Dimensions: L = 30mm; W = 11mm (approximately).
Alloy	Eutectoid to hypereutectoid steel
Technology	Quench hardened
Lab number of sample	DEV1069
Sample location	Empa (Marianne Senn)
Responsible institution	Office de la Culture, Porrentruy, Jura
Date and aim of sampling	2000, metallography and chemical composition of the metal

Complementary information

Nothing to report.

Analyses and results

Analyses performed:
Metallography (nital etched), Vickers hardness testing, LA-ICP-MS, SEM/EDX.

Non invasive analysis

Metal

The remaining metal is a eutectoid to hypereutectoid steel (C 0.8-1 mass%) with cracks and porosity (Figs. 5 and 6). Its composition is given in Table 1 (except Fe). Of the trace elements, only Ni reaches a medium concentration (Ni 0.07 mass%). The composition and the high Ni/Co ratio are atypical for iron worked in smithies of the settlement at Develier-Courtételle JU, CH. The few slag inclusions and their chemical composition confirm this (Table 2). The slag is rich in CaO whereas the pisolithic or bean ore worked in the Central Jura smelting district is rich in Al_2O_3. The slag inclusions cannot be smithing slag because the artefact is formed from one piece of metal. The slag probably incorporated while refining the bloom or during the bloom smelting process. The ore from which this metal was smelted cannot yet be identified. Intergranular corrosion has developed, penetrating into the metal structure (Fig. 6). The etched metal shows a dominant martensitic structure, typical of quench hardened steel (Figs. 5, 7, 8 and 9). The characteristic forms of the martensite are non-organised needles (Fig. 8). In restricted areas bainite nodules occur (Figs. 7 and 8). In the centre of the point, hypereutectoid steel is composed of cementite in the grain boundaries and in needle like shapes within the grains (Fig. 10). Hypereutectoid regions also occur in the centre of the shaft and on its left surface (Fig. 5). Here the hypereutectoid steel is composed of martensite and cementite in the grain boundaries and in needle form within the grains (Figs. 6 and 9). Nevertheless, most of the chisel is dominated by eutectoid steel. The average hardness of the quench hardened eutectoid steel is about HV1 690. In the bainite nodules the average hardness drops to about HV1 350, the martensite has an average hardness of HV1 750.

Elements	V	Cr	Mn	P	Co	Ni	Cu	As	Ag	Ni/Co	C* mass%
Median mg/kg	2	20	2	300	40	680	100	190	<	17	0.8-1
Detection limit mg/kg	1	8	1	70	1	1	1	3	0.2	1	6
RSD %	50	36	15	21	5	13	55	10	-		

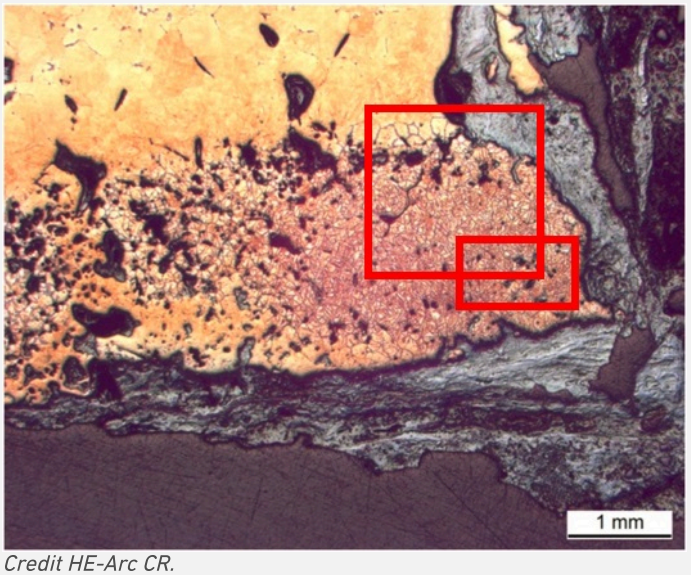
*visually estimated

Table 1: Chemical composition of the metal. Method of analysis: LA-ICP-MS, Laboratory of Analytical Chemistry, Empa (for details see Devos et al. 2000).

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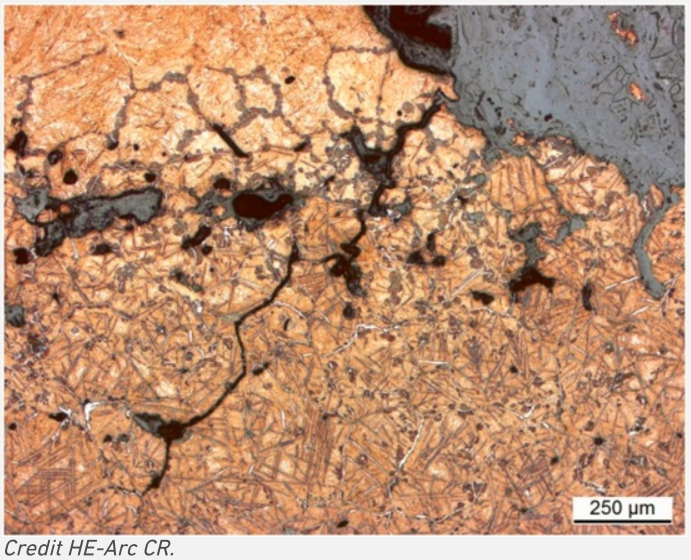
Structure	Location	Na_2O	MgO	Al_2O_3	SiO_2	K_2O	CaO	TiO_2	FeO	Total	SiO_2/Al_2O_3
Glass and?	Tip	0.6	2.4	7.8	50	3.9	15	<	20	100	6.4
Glass and?	Tip	0.6	2.5	8.9	65	4.7	18	0.7	4.2	104	7.3
Glass with needles	Middle	0.7	2.6	10	59	4.6	21	0.7	2.5	101	5.9
Glass with needles	Middle	0.7	2.5	8.7	58	4.7	21	0.6	3.0	100	6.7
Glass with needles	Middle	0.8	2.3	8.5	51	4.1	24	0.7	3.6	95	6.0

Table 2: Chemical composition of the slag inclusions (mass%). Method of analysis: SEM/EDS, Laboratory of Analytical Chemistry, Empa.



Credit HE-Arc CR.

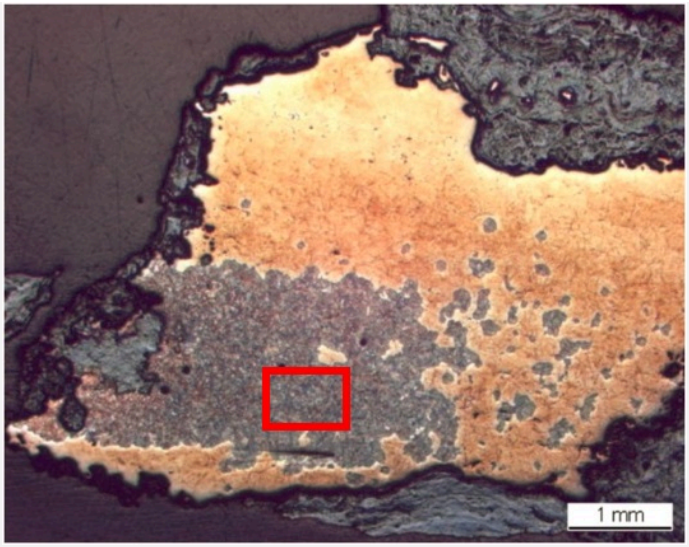
Fig. 5: Micrograph of the metal sample from Fig. 3 (inverted picture, detail), etched, bright field. The metal contains porosity and cracks (in black). The micrographs of Figs. 6 and 9 are marked by the top and bottom rectangles respectively,



Credit HE-Arc CR.

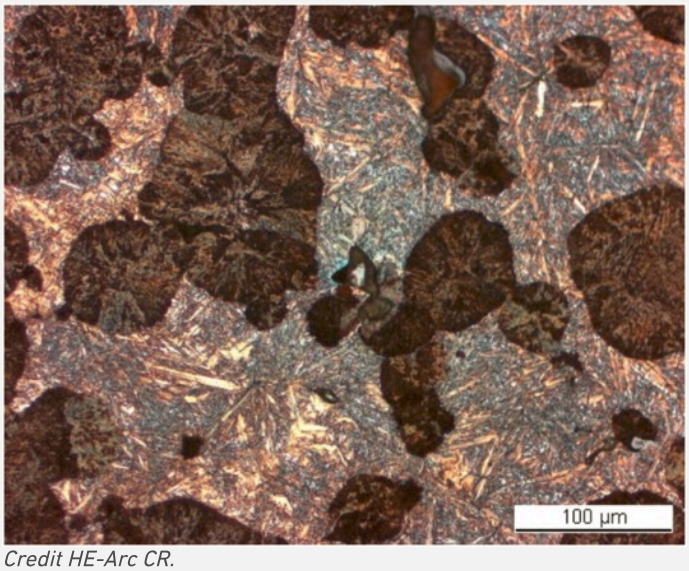
Fig. 6: Micrograph of the metal sample, detail from Fig. 5, etched, bright field. We observe martensite (in orange), cementite (in grey), porosity and intergranular cracks (in black) and corrosion,

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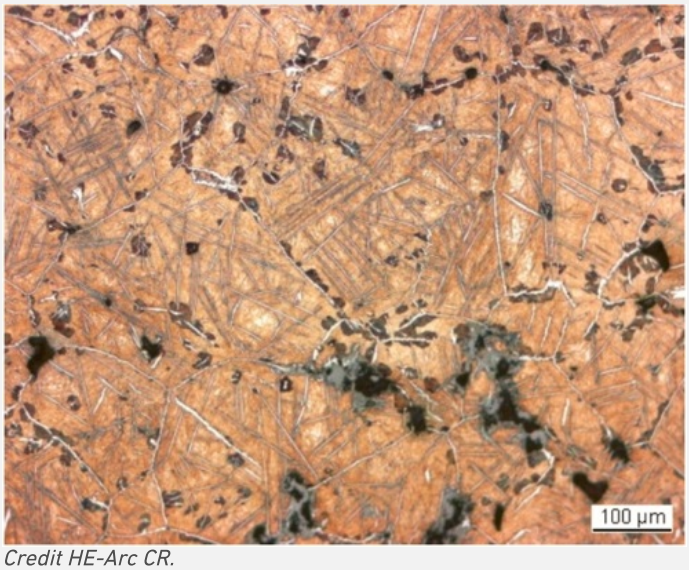
Credit HE-Arc CR.

Fig. 7: Micrograph of the metal sample from Fig. 3 (inverted picture, detail), etched, bright field. We observe the metal with martensite (orange) and bainite nodules (blue). The micrograph of Fig. 10 is marked by a rectangle,



Credit HE-Arc CR.

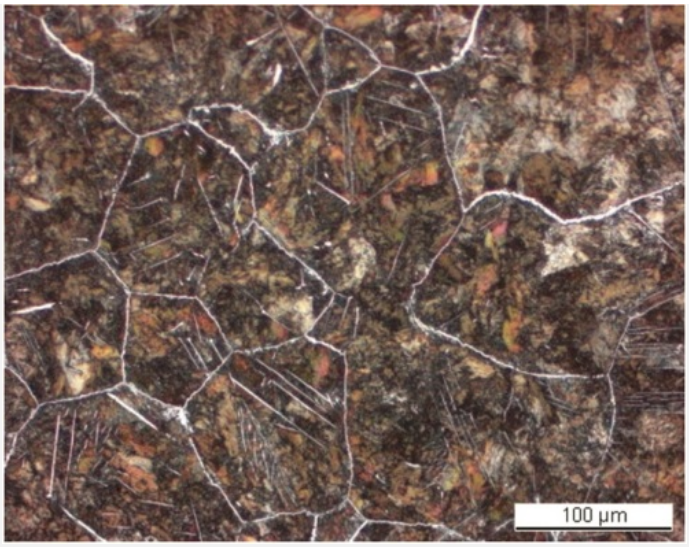
Fig. 8: Micrograph of the metal sample from Fig. 3 (detail), etched, bright field. We observe dark bainite nodules combined with martensite needles,



Credit HE-Arc CR.

Fig. 9: Micrograph of the metal sample, detail from Fig. 5, etched, bright field. Hypereutectoid steel with a structure of cementite in white in the grain boundaries and in needle form combined with martensite (pink). Some bainite occurs around the grain boundaries,

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Credit HE-Arc CR.

Fig. 10: Micrograph of the metal sample, detail from Fig. 7, etched, bright field. Hypereutectoid steel with a structure of cementite in white in the grain boundaries and in needle form combined with bainite (dark),

Microstructure	Martensitic structure (non organised needles) + bainite nodules
First metal element	Fe
Other metal elements	C

Complementary information

Nothing to report.

Corrosion layers

The metal is heavily corroded and the thickness of the corrosion crust is irregular, but averages about 0.5mm. The corrosion products enclose different ghost structures: the corrosion crust has preserved the shape of polygonal grains (Figs. 11 and 13) and cementite needles (Fig. 12). Under polarised light, the corrosion products at the metal - corrosion crust interface are dark-red and form an inner S-rich layer (CP3, Figs. 11 and 13). Cl occurs locally (CP3) and C is present in layers CP2 and CP3 (Fig. 13). The outer layer of the corrosion products (CP1) is orange and contains more O than the inner layer CP3 (Figs. 11 and 14).

Elements	O	S	Cl	Fe	Total
Inner red/light-grey layer (CP3)	27	1.2	1.3	72	103
Inner dark-red layer (average of 2 similar analyses) (CP2)	31	0.9	<	66	99
Inner orange grain zone	36	<	<	63	101
Outer orange sub-layer (average of 2 similar analyses) (CP1)	35	<	<	63	98

Table 3: Chemical composition (mass %) of the corrosion crust (from Figs. 11-14). Method of analysis: SEM/EDS, Laboratory of Analytical Chemistry, Empa.

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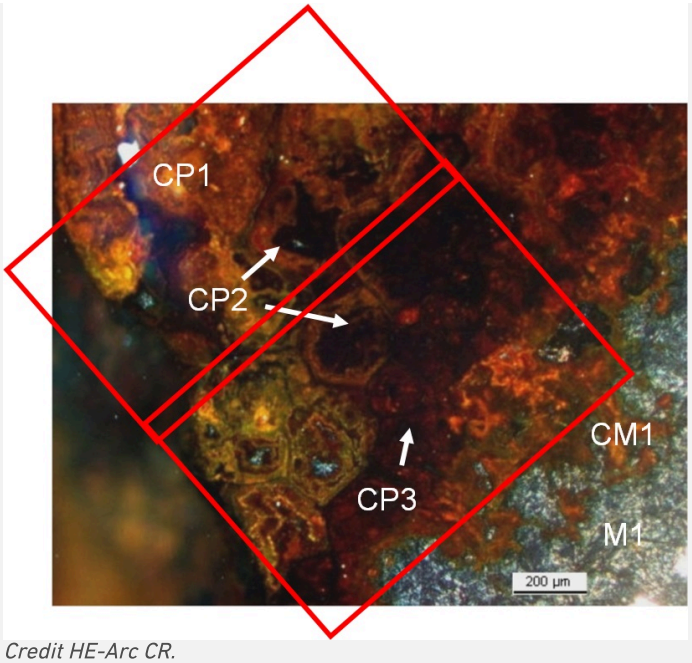


Fig. 11: Micrograph showing the metal - corrosion crust interface from Fig. 3 (rotated by 270°, detail) and corresponding to the stratigraphy of Fig. 4, unetched, polarised light. Polygonal grains are visible as ghost structure in the corrosion crust. The selected areas for elemental chemical distribution (Figs. 13 and 14) are marked by two rectangles (bottom: Fig. 13 and top: Fig. 14),

Credit HE-Arc CR.

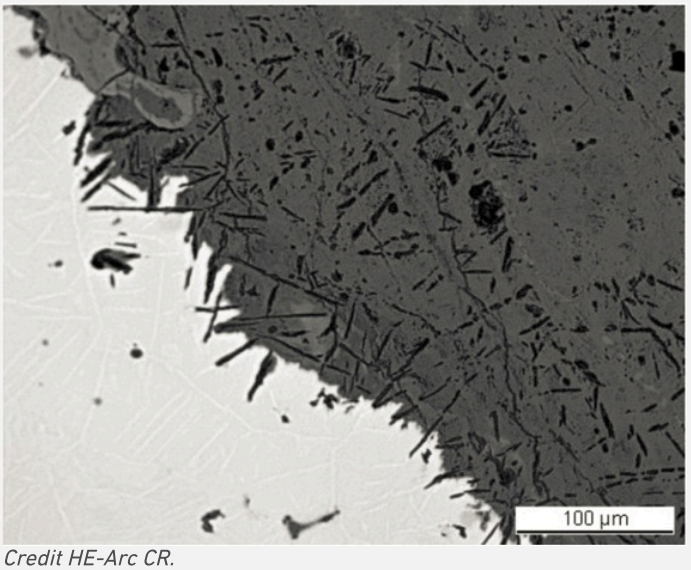


Fig. 12: Micrograph showing the metal - corrosion crust interface from Fig. 3 (detail), unetched, bright field. The cementite is visible as white needles in the metal and its ghost structure as black needles in the corrosion crust,

Credit HE-Arc CR.

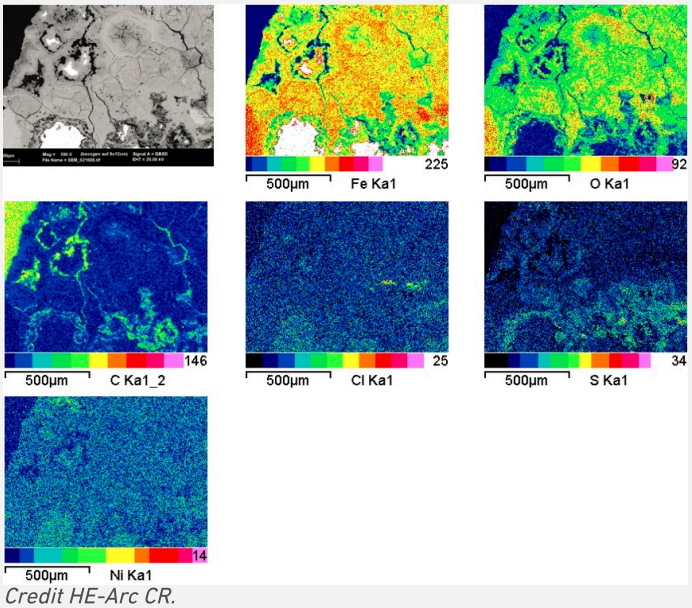


Fig. 13: SEM image, BSE-mode, and elemental chemical distribution of the selected area from Fig. 11 (rotated by 45°). Method of examination: SEM/EDS, Laboratory of Analytical Chemistry, Empa,

Credit HE-Arc CR.

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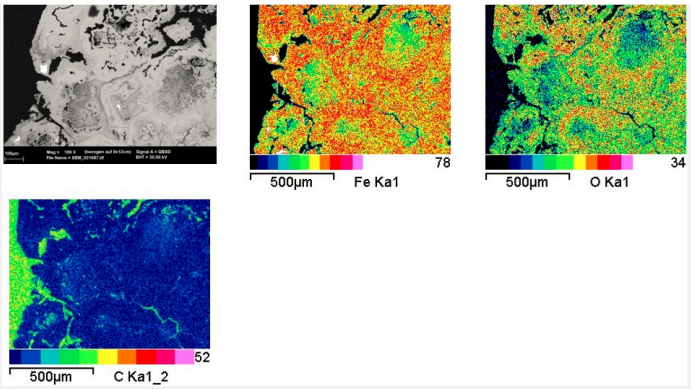


Fig. 14: SEM image, BSE-mode, and elemental chemical distribution of the selected area from Fig. 11 (rotated by 45°). Method of examination: SEM/EDS, Laboratory of Analytical Chemistry, Empa,

Credit HE-Arc CR.

Corrosion form	Uniform - intergranular
Corrosion type	?

Complementary information

Nothing to report.

MiCorr stratigraphy(ies) – CS

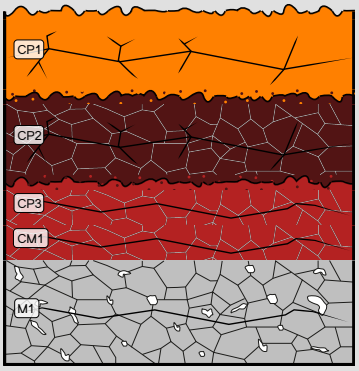


Fig. 4: Stratigraphic representation of the chisel point in cross-section (dark field) using the MiCorr application. The characteristics of the strata are only accessible by clicking on the drawing that redirects you to the search tool by stratigraphy representation. This representation can be compared to Fig. 11, Credit HE-Arc CR.

Synthesis of the binocular / cross-section examination of the corrosion structure

Corrected stratigraphic representation: none.

Conclusion

The tool is made of hard, eutectoid to hypereutectoid steel and was, as a last manufacturing step, entirely quenched. In this state it is too hard and brittle to be used. It is likely that the quenched hypereutectoid structure provoked the breakage of the tool when it was first used. The absence of a tempering process at the end is unclear. The chemical composition of the metal and the slag inclusions indicate that the artefact was imported to the hamlet of Develier-Courtételle JU. The corrosion has, in areas, replaced the metal structure retaining a ghost structure; the outer layer has been mostly removed during surface cleaning. Extensive pitting corrosion has occurred, typical for steel with an elevated C content. The presence of S in the inner layer can be explained by putrefaction processes in the soil. Chlorides are still present at the metal-corrosion interface as no desalination process was attempted. Iron carbonates are present in the centre of a corrosion pit. The corrosion is of terrestrial type.

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References

References on object and sample

References object

1. Eschenlohr, L., Friedli, V., Robert-Charrue Linder, C., Senn, M. (2007) Develier-Courtételle. Un habitat mérovingien. Métallurgie du fer et mobilier métallique. Cahier d'archéologie jurassienne 14 (Porrentruy), 314-315.

References sample

2. Eschenlohr, L., Friedli, V., Robert-Charrue Linder, C., Senn, M. (2007) Develier-Courtételle. Un habitat mérovingien. Métallurgie du fer et mobilier métallique. Cahier d'archéologie jurassienne 14 (Porrentruy), 275.

References on analytic methods and interpretation

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