



COFFEE FILTER - AL ALLOY - MODERN TIMES - FRANCE

Artefact name Coffee filter

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Url /artefacts/711/

▼ The object



Fig. 1: Front and back sides of a coffee filter,

▼ Description and visual observation

Description of the artefact Coffee filter with traces of use (deposits, deformation) and presence of local filiform corrosion

(Fig. 1). Dimensions: ø.ext. = 10cm.

Type of artefact Household implement

Origin Château de Germolles, Mellecey, Bourgogne, France

Recovering date Unknown

Chronology category Modern Times

chronology tpq 1960 A.D. 🗸

chronology taq 2000 A.D. ∨

Chronology comment 20th century

Burial conditions / Outdoor atmosphere environment

Artefact location Château de Germolles, Mellecey, Bourgogne

Owner Château de Germolles, Mellecey, Bourgogne Inv. number None

Recorded conservation data Not conserved

Complementary information

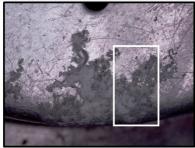
Nothing to report.





Fig. 2: Detail of the front and back sides of the coffee filter showing the location of Fig. 3,





Credit HE-Arc CR, J.Schröter.

Fig. 3: Detail of the back side of the coffee filter showing the location of the sampling area,

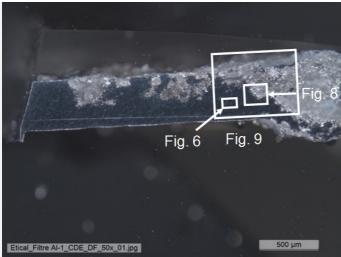
Stratigraphic representation: none.

★ MiCorr stratigraphy(ies) - Bi

Fig. 4: Micrograph of the cross-section of the fragment sampled from the coffee filter showing the location of Figs. 6, 8 and 9, dark field,







Credit HE-Arc CR, J.Schröter.

Description of sample Sample cut from the back side of the coffee filter (Fig. 3).

Alloy Al Alloy

Technology Annealed after (hot) rolling

Lab number of sample None

HE-Arc CR, Neuchâtel, Neuchâtel Sample location

Responsible institution HE-Arc CR, Neuchâtel, Neuchâtel

Date and aim of sampling 2017, EtICAL project (Study of corrosion forms of aluminium alloys in Swiss collections)

Complementary information

A second sample was taken and gave similar results.

Analyses performed: Metallography, SEM/EDS.

The metal is a relatively pure aluminium alloy with numerous elongated inclusions (Fig. 6). From their chemical composition they can be interpreted as Al3Fe intermetallic compounds (Fig. 7). Inter- and transgranular corrosion has





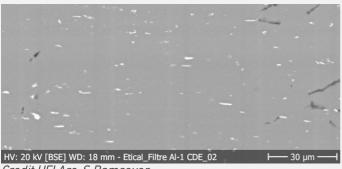


Fig. 6: SEM image of the metal sample from Fig. 4 (detail), BSE-mode. We observe the presence of numerous elongated

Credit HEI Arc, S.Ramseyer.

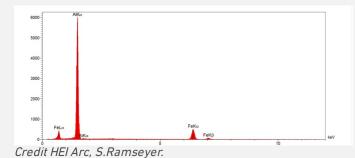


Fig. 7: EDS spectrum of the elongated inclusions of Fig. 6,

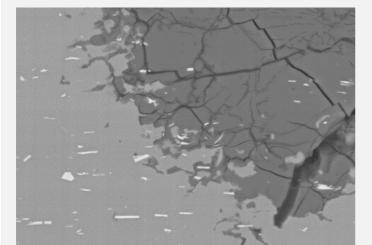


Fig. 8: SEM image of the metal sample from Fig. 4 (detail), BSE-mode. Extensive inter- and transgranular corrosion has developed within the metal,

Credit HEI Arc, S.Ramseyer.

HV: 20 kV [BSE] WD: 18 mm - Etical_Filtre Al-1

Microstructure Recrystallized structure (polygonal grains)

First metal element Αl

Other metal elements Fe

Complementary information

Nothing to report.

The average thickness of the corrosion layer (CP1) is about 50 micrometers, but may be thinner or thicker depending on the area. Intergranular corrosion developed locally to extend through the entire thickness of the metal. EDS-SEM





analysis indicates that the metal is, as expected, covered by a layer rich in Al and O containing chlorides (red spots on Fig. 9) and surprisingly Na (Figs. 10 and 11). Chlorides do not seem to form active corrosion. A new examination carried out after 4 months (Fig. 12) shows new forms of alteration with a local enrichment of Na, C and O (Na2CO3?).

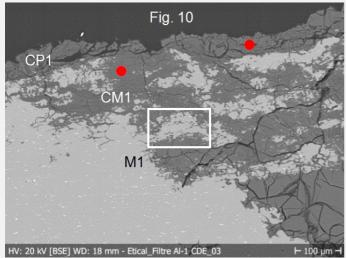


Fig. 9: SEM picture (detail of Fig. 4 with location (red spots) of EDS analyses of Fig. 10), BSE-mode. From bottom to top left: the metal (M1) in light grey, the corroded metal (CM1) and CP1. The area selected for elemental chemical distributions (Figs. 11 and 12) is marked by a rectangle,

Credit HEI Arc, S.Ramseyer.



Fig. 10: EDS spectrum of red spots in Fig. 9,

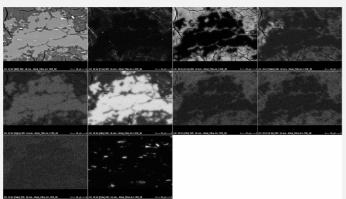
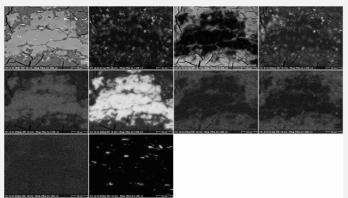


Fig. 11: SEM image, BSE-mode, and elemental chemical distribution of the selected area of Fig. 9. Method of examination: SEM-EDS, Lab of Electronic Microscopy and Microanalysis, IMA (Néode), HEI Arc,

Credit HEI Arc, S.Ramseyer.



HEI Arc, S. Ramseyer.

Credit HEI Arc, S. Ramseyer.





Corrosion form Multiform - intergranular

Corrosion type None

Complementary information

Nothing to report.

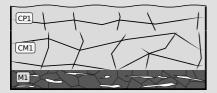


Fig. 5: Stratigraphic representation of the fragment sampled from the coffee filter in cross-section (dark field) using the MiCorr application. The characteristics of the strata are only accessible by clicking on the drawing that redirects you to the search tool by stratigraphy representation. This representation can be compared to Fig. 9, Credit HE-Arc CR, C.Degrigny.

♥ Synthesis of the binocular / cross-section examination of the corrosion structure

Corrected stratigraphic representation: none.

♥ Conclusion

This aluminium alloy has a composition similar to a primary aluminium with an Al content between 99 and 99.8 mass%. The main impurity is Fe forming intermetallic (Al3Fe) inclusions. The metal was stamped and punctured. It is covered by a relatively thick corrosion layer (probably aluminium oxide) which developed under superficial filiform corrosion. Extensive intergranular corrosion has developed locally within the metal. Chlorides have been identified but the progress of the corrosion might be due to Na, C and O-rich compounds (Na2CO3?).

▼ References

References on object and sample

References object

- 1. Degrigny, C. (2018) Etude, identification des objets en aluminium patrimoniaux et classification de leurs formes de corrosion - projet EtICAL, rapport interne HE-Arc CR.
- 2. Degrigny, C. and Schröter, J. (2019) Aluminium alloys in Swiss public collections: identification and development of diagnostic tools to assess their condition, in Metal 2019, proceedings of the ICOM-CC Metal WG interim meeting, eds. C. Chemello, L. Brambilla, E. Joseph, Neuchâtel (Switzerland), 408-415.

References sample

3. Degrigny, C. (2018) Etude, identification des objets en aluminium patrimoniaux et classification de leurs formes de corrosion - projet EtICAL, rapport interne HE-Arc CR.





References on analytic methods and interpretation	



