



CORRODED METAL SHEET FN 655.13.01 - FE ALLOY - MODERN TIMES - SWITZERLAND

Artefact name Corroded metal sheet FN 655.13.01

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Neuchâtel, Switzerland)

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▼ The object



Credit HE-Arc CR.

Fig. 1: Metal sheet (after Senn Bischofberger 2005, 137),

▼ Description and visual observation

Description of the artefactMetal (iron-based) sheet with middle rib. The shape is no longer discernible (Fig. 1). The metal

is covered by a thick corrosion crust. Dimensions: L = 4.5cm; WT = 35g.

Type of artefact Metal sheet

Origin Steinmöri, Neftenbach / Dorf Neftenbach, Zurich, Switzerland

Recovering date Excavation of the Roman villa, 1986-1990, phase S2, StbII.2 (2nd/3rd century AD)

Chronology category Modern Times

chronology tpq 1900 A.D. ▶

chronology taq 2000 A.D. •

Chronology comment Product of the 20th century AD introduced accidentally into a Roman layer

Burial conditions / environment

Soil

Artefact location

Kantonsarchäologie, Dübendorf, Zurich

Owner

Kantonsarchäologie, Dübendorf, Zurich

Inv. number

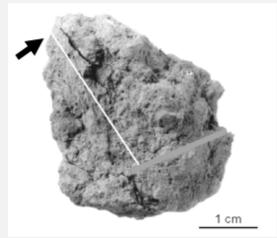
FN 655.13.01

Recorded conservation data

N/A

Complementary information

None.



Credit HE-Arc CR.

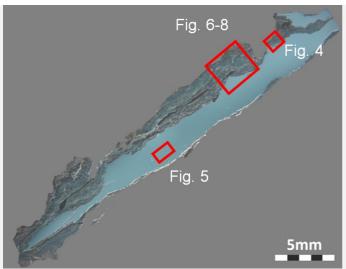
Fig. 2: Location of sampling area,

st $\,$ Binocular observation and representation of the corrosion structure

None.

★ MiCorr stratigraphy(ies) - Bi

Fig. 3: Micrograph of the cross-section of the metal sheet showing the location of Figs. 4 to 8,



Credit HE-Arc CR.

Description of sampleA longitudinal cut has been made through the sheet (Fig. 2). Large parts of the corrosion crust

fell off during sampling (Fig. 3).

Alloy Fe Alloy

Technology Forged with final cold work

Lab number of sample NEF 655

Sample location HE-Arc CR, Neuchâtel, Neuchâtel

Responsible institution Kantonsarchäologie, Dübendorf, Zurich

Date and aim of sampling 2000, metallography

Complementary information

None.

★ Analyses and results

Analyses performed:

Metallography (nital etched), Vickers hardness testing, LA-ICP-MS, SEM/EDS.

Non invasive analysis

None.

× Meta

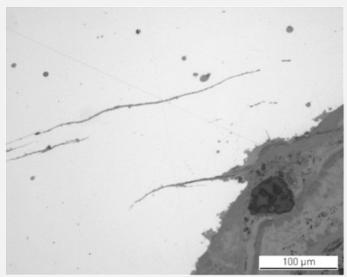
The remaining metal is an iron containing elevated (more than 1g/kg) concentrations of Co and Ni (Table 1), small round slag inclusions and cracks resulting from corrosion and deformation (Fig. 4). The composition of the round, elongated slag inclusions is similar to wüstite-FeO (Table 2). Iron reduced in the direct smelting process never contains such pure compounds and only FeO inclusions. In modern steels FeO occurs in low carbon alloys and Armcoiron (Schumann 1991, 474). After etching, the cross-section shows a heavily deformed ferritic microstructure from cold working (Fig. 6). The cracks have the same orientation as the deformation. The average hardness of the metal is HV1 195.

Elements	Al	Ti	٧	Cr	Mn	Р	Со	Ni	Cu	As	Мо	Ag	Sn	Sb	W	Ni/Co
Median, mg/kg	<	<	<	<	20	300	1100	1200	800	200	<	<	60	10	<	1.1
RSD %	-	-	-	-	3	17	6	5	35	15	13	-	35	45	-	4
Detection limit mg/kg	6	9	19	17	2	77	1	4	2	3	3	1	1	1	3	-

Table 1: Chemical composition of the metal sheet. Method of analysis: LA-ICP-MS, Lab of Inorganic Chemistry, ETH.

Elements	0	Fe	Total
Round inclusion 1	21	75	96
Round inclusion 2	20	74	95
Round inclusion 3	20	75	96
Round inclusion 4	20	73	93

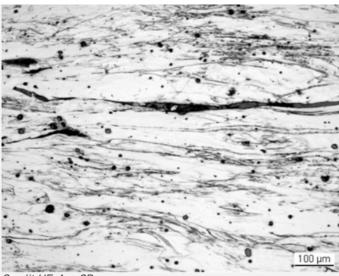
Table 2: Chemical composition (mass %) of some of the round inclusions of Fig. 4. Method of analysis: SEM/EDS, Laboratory of Analytical Chemistry, Empa.



Credit HE-Arc CR.

Fig. 4: Micrograph of metal sample from Fig. 3 (inverted picture, detail), unetched, bright field. In white the metal with round slag inclusions as well as cracks filled with corrosion products,

Fig. 5: Micrograph of metal sample from Fig. 3 (detail), etched, bright field. The heavily cold worked metal shows elongated grains with nearly invisible grain boundaries and cracks parallel to the grains,



Credit HE-Arc CR.

Microstructure Heavily deformed ferritic microstructure

First metal element Fe

Other metal elements Co, Ni

Complementary information

None.

▼ Corrosion layers

The corrosion is massive and large parts of the outer corrosion layers have been lost during the sample preparation. The heavily cracked corrosion crust represents about one third of the thickness of the sample (Fig. 3) and is located on one side of the metal (the rest having been lost during the cutting process). It does not show well defined layers. Despite this, three areas can be distinguished (Figs. 6 and 7). Adhering to the metal (area 1), we find an orange-red-brown layer enriched in Cl (CP3 in Fig. 9, Fig. 8 and Table 3). The dark or violet middle area 2 is richer in Fe (CP2 in Fig. 9). The outer area 3 is red-brown (CP1 in Fig. 9) and strongly contaminated by soil material (rock fragment inclusion and elements like Si, Al, P etc.). This layer is enriched in O (Fig. 9).

Elements	Location	0	Na	Al	Si	Р	Cl	Fe	Cu	Мо	Total
	Dark-brown	31	<	<	<	<	<	67	<	<	99
	Dark-brown	35	<	<	<	<	0.7	64	<	<	100
Area 1 (CP3)	Red-brown	38	<	<	<	<	2.2	64	<	0.6	104
	Red-brown	36	<	<	<	<	3.3	67	<	<	106
	Red-brown	39	<	<	<	<	1.3	63	<	<	104
	Orange-brown	35	1.3	<	<	<	<	61	<	<	98
Area 2 (CP2)	Orange-brown	31	<	<	<	<	0.7	63	<	<	95
	Violet-brown	30	<	<	<	<	<	75	0.8	1.0	108
	Brown	30	<	<	<	<	<	77	<	0.6	108
	Violet-brown	35	<	<	<	<	<	68	<	<	103

	Yellow	33	<	<	<	<	<	67	<	<	100
	Violet-brown	32	<	<	<	<	<	74	<	<	106
Area 3 (CP1)	Quartz inclusion	54	<	<	52	<	<	<	<	<	106
	Mixture brown	42	<	<	1.6	0.7	<	58	<	<	102
	Mixture brown	39	<	1.9	4.3	<	<	56	<	<	102
	Mixture brown	35	<	<	0.9	<	<	59	<	<	95

Table 3: Chemical composition (mass %) of the corrosion products from near the metal (area 1, CP3) to the outer surface (area 3, CP1). Method of analysis: SEM/EDS, Laboratory of Analytical Chemistry, Empa.

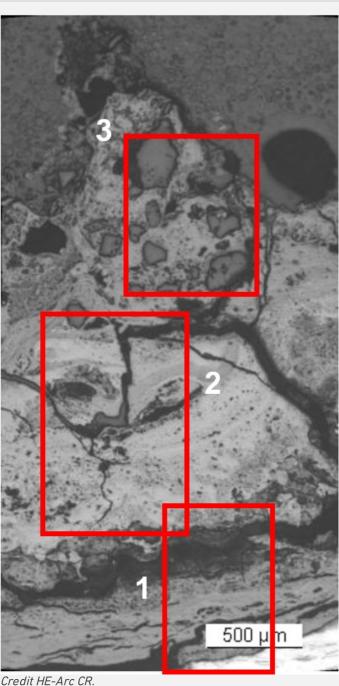
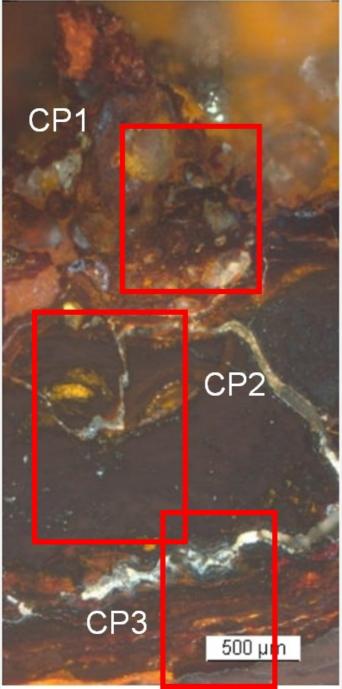


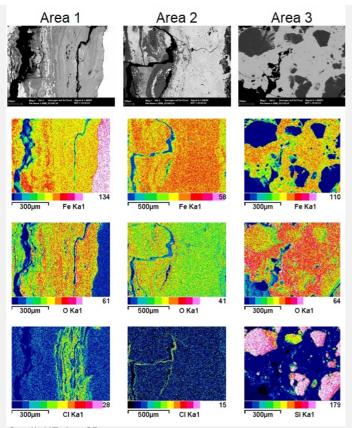
Fig. 6: Micrograph of the corrosion layer from Fig. 3 showing the metal - corrosion crust interface (detail, rotated by 45°) and corresponding to the stratigraphy of Fig. 9, unetched, bright field. The areas 1 (for CP3), 2 (for CP2) and 3 (for CP3) selected for elemental chemical distribution (Fig. 8) are marked by red rectangles,

Fig. 7: Micrograph (same as Fig.6) and corresponding to the stratigraphy of Fig. 9, polarised light. The inner layer (CP3) is red-brown, area 2 is brown-violet (CP2) and the outer brown area contains rock fragment inclusions (CP1),



Credit HE-Arc CR.

Fig. 8: SEM images, BSE-mode, and elemental chemical distribution of the selected areas from Figs. 6 and 7 (reversed picture rotated by 270°, details). First column: mapping on area 1 (CP3) where an inner layer enriched in Cl appears near the metal surface. 2nd column: area 2 (CP2) where iron oxides prevail. 3rd column: area 3 (CP1) where the outer 0-rich layer includes rock fragments rich in Si. Method of examination: SEM/EDS, Lab. Anal. Chem. Empa,



Credit HE-Arc CR.

Corrosion form Uniform - transgranular

Corrosion type Unknown

Complementary information

None.

MiCorr stratigraphy(ies) – CS

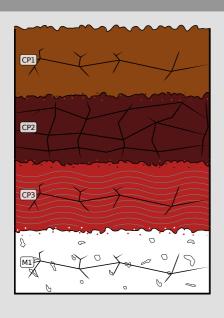


Fig. 9: Stratigraphic representation of the iron-based sheet in cross-section (dark field) using the MiCorr application. The characteristics of the strata are only accessible by clicking on the drawing that redirects you to the search tool by stratigraphy representation. This representation can be compared to Figs. 7, Credit HE-Arc CR.

▼ Synthesis of the binocular / cross-section examination of the corrosion structure
None.
★ Conclusion
This iron-based sheet is entirely cold worked and hard compared to an annealed metal. The corrosion is massive and masks the shape of the object. The presence of Cl adjacent to the metal surface indicates that the corrosion front is potentially active. The very thick top corrosion layers may have slowed down the corrosion. The object was

▼ References

References on object and sample

References object

1. Rychener, J. (1999) Der römische Gutshof in Neftenbach. Katalog, Tafeln und Tabellen. Monographien der Kantonsarchäologie Zürich 31/2 (Zürich und Egg), 138.

mentioned as being Roman in Senn Bischofberger 2005, however the chemical composition of the slag inclusions shows that it is a product of the 20^{th} century AD which has been accidentally introduced into a Roman layer.

References sample

2. Senn Bischofberger, M. (2005) Das Schmiedehandwerk im nordalpinen Raum von der Eisenzeit bis ins frühe Mittelalter. Internationale Archäologie, Naturwissenschaft und Technologie Bd. 5, (Rahden/Westf.), 137-138.

References on analytic methods and interpretation

3. Schumann, H. (1991) Metallographie. Leipzig.