



# FITTING ON BACK PANEL OF A MILITARY CARRIAGE – THOMAS STEEL – MODERN TIMES – SWITZERLAND

Artefact name Fitting on back panel of a military carriage

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Url /artefacts/899/

#### ▼ The object





Credit HE-Arc CR, A.Tarchini.

Fig. 1: Fitting on back panel of a military carriage,

# imes Description and visual observation

**Description of the artefact** Fitting on back panel, first exposed outdoors, later indoors (Fig. 1). Uniform corrosion and

pitting corrosion are visible.

Type of artefact Military carriage

Origin Swiss Army, Thun, Bern, Switzerland

**Recovering date**Built by Konstruktions-Werkstätte, 1918

Chronology category Modern Times

chronology tpq 1918 A.D. ✓

chronology taq

**Chronology comment** 

Burial conditions / environment

Outdoor to indoor atmosphere

Artefact location Historical Swiss Army Material Foundation, Burgdorf, Bern

**Owner** Historical Swiss Army Material Foundation, Burgdorf, Bern

Inv. number None.

Recorded conservation data N/A

### **Complementary information**

None.



Fig. 2: Location of sampling area,

Credit HE-Arc CR.

#### Binocular observation and representation of the corrosion structure

None.

# ★ MiCorr stratigraphy(ies) - Bi

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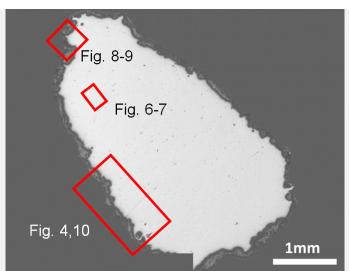


Fig. 3: Micrograph of the cross-section of the sample taken from the fitting showing the location of Figs. 4 and 6 to 10,

Credit HE-Arc CR.

**Description of sample**This sample is a cut from the corner of one of the two fittings on the back panel (Fig. 2). The

metal is covered by a thin corrosion layer (Fig. 3).

Alloy Thomas steel

**Technology** Piled from several strips, hot rolled and annealed

Lab number of sample POINT-Fe2

Sample location HE-Arc CR, Neuchâtel, Neuchâtel

Responsible institution Historical Swiss Army Material Foundation, Burgdorf, Bern

Date and aim of sampling 05/2009 metallography

#### Complementary information

None.

## ★ Analyses and results

#### Analyses performed:

Metallography (nital etched after etching with Oberhoffer's reagent), Vickers hardness testing, LA-ICP-MS, SEM/EDS.

#### Non invasive analysis

None.

The remaining metal is a Mn-rich soft steel (C content around 0.1 mass%) containing manganese sulphide inclusions with a varying Fe content (Tables 1 and 2). The numerous inclusions form parallel rows (Fig. 4). This orientation is typical for hot rolled metal. After etching with Oberhoffer's reagent, three main welding seams (P-rich) become visible (Fig. 5). Near the surface they are also outlined by corrosion (Figs. 4 and 10). After nital etching, the metal shows a ferritic structure with tertiary cementite and lamellar pearlite at the grain boundaries (Figs. 6 and 7). The grains are small with an ASTM grain size of 10 and are recrystallized due to annealing after hot rolling. The average hardness of the metal is HV1 165. The hardness is slightly high for such a structure and this is due to the Mn content of the metal. The chemical composition, especially the Mn content and the presence of carbo-nitrides (not analysed here), is typical for Thomas steel.

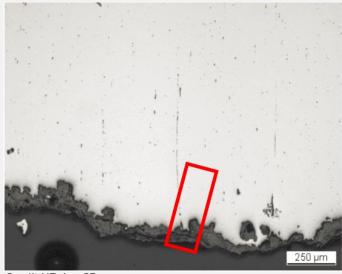
Elements	Ni/Co	Al	Р	Ti	٧	Cr	Mn	Со	Ni	Cu	As	Мо	Ag	Sn	Sb	W	C* mass%
Median (mg/kg)	2.4	<	300	<	<	140	3600	200	480	140	700	10	<	10	10	<	<0.1
Detection Limit (mg/kg)		5	82	10	2	13	2	1	3	1	3	3	1	1	1	4	
RSD %	1	-	8	-	-	3	7	2	1	6	3	7	-	7	8	-	

\*visually estimated

Table 1: Chemical composition of the metal. Method of analysis: LA-ICP-MS. Lab Inorganic Chemistry, ETH.

Elements	S	Mn	Fe	Cu	Total
Inclusions	26	46	31	2.2	105

Table 2: Chemical composition (mass %) of the inclusions. Method of analysis: SEM/EDS, Laboratory of Analytical Chemistry, Empa.



Credit HE-Arc CR.

Fig. 4: Micrograph of the metal sample from Fig. 3 (rotated by 270°, detail), unetched, bright field. We observe the metal including welding seams outlined by corrosion products and MnS inclusions. The area selected for elemental chemical distribution (Fig. 10) is marked by a red rectangle,

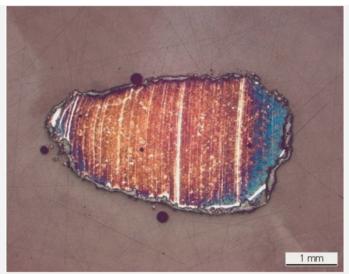


Fig. 5: Micrograph of the metal sample from Fig. 3, etched with Oberhoffer's reagent, bright field. We observe three Prich welding seams in white,

Credit HE-Arc CR.

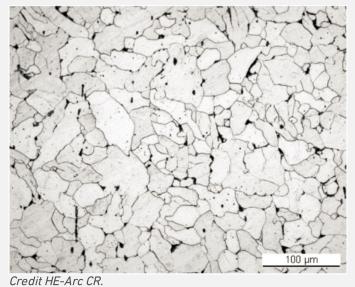


Fig. 6: Micrograph of the metal sample, nital etched, bright field. We observe tertiary cementite (black) on the grain boundaries.

CIEUIL HE-AIL CK.

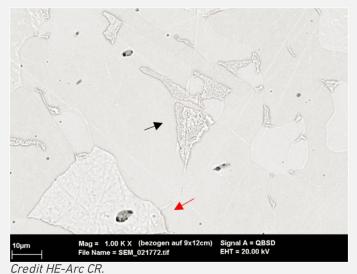


Fig. 7: SEM image of the metal sample from Fig. 3 (detail), BSE-mode, nital etched, bright field. The white ferrite grains contain lamellar pearlite (black arrow) and lenticular grey MnS inclusions (red arrow) in between the grain boundaries,

CIEUIL NE-AIL CK.

Microstructure Recrystallized grain structure with tertiary cementite

First metal element Fe

Other metal elements C, Mn

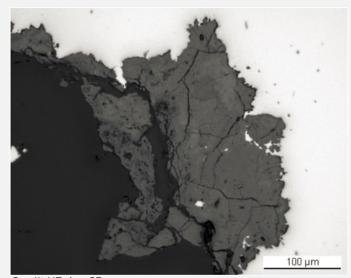
None.

#### ▼ Corrosion layers

The average thickness of the corrosion products is about  $80\mu m$  (Figs. 4 and 8). In bright field they appear grey, marbled and heavily cracked (Fig. 8). Under polarised light, the corrosion products appear orange to dark-brown (Fig. 9). At the metal - corrosion products interface they are dark-brown (CP3). The middle part (CP2) is dark-red and the outer part is bright orange (CP1). The elemental mapping of the corrosion layers shows no distinctive stratification, but areas near the metal - corrosion crust interface (CP3) as well as the top surface of the corrosion layer (CP1) seem to have a lower 0 content (Fig. 10). The 0 content indicates the presence of iron hydroxides (Table 3). Soil materials (such as rock fragments and dust) are found in the welding seams near the surface.

Location	Elements	0		Mn	Fe	Total
In welding seam		34	<	<	67	102
Inner corrosion laye	er (CP3)	38	0.7	0.8	66	106

Table 3: Chemical composition (mass %) of the corrosion layer (from Fig. 10). Method of analysis: SEM/EDS, Laboratory of Analytical Chemistry, Empa.



Credit HE-Arc CR.

Fig. 8: Micrograph showing the metal - corrosion crust interface from Fig. 3 (rotated by 270°, detail), unetched, bright field. We observe in white the metal, in grey the corrosion products and in black the resin,

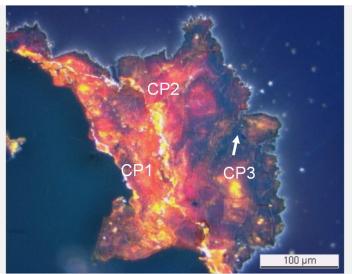


Fig. 9: Micrograph showing the metal - corrosion crust (same as Fig. 8) and corresponding to the stratigraphy of Fig. 11, unetched, polarised light. The corrosion products are darkbrown at the metal - corrosion crust interface and redorange on the outside,

Credit HE-Arc CR.

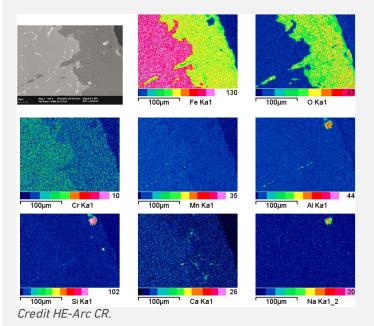


Fig. 10: SEM image, BSE-mode, and elemental chemical distribution of the selected area from Fig. 4 (rotated by 270°, detail). Method of examination: SEM/EDS, Laboratory of Analytical Chemistry, Empa.

Corrosion form Uniform - transgranular

Corrosion type Unknown

#### Complementary information

Nothing to report.

### ★ MiCorr stratigraphy(ies) – CS

Fig. 11: Stratigraphic representation of the sample taken from the fitting in cross-section (dark field) using the MiCorr application. The characteristics of the strata are only accessible by clicking on the drawing that redirects you to the search tool by stratigraphy representation. This representation showing for the metal part a welding seam (M2) can be compared to Fig. 9, Credit HE-Arc CR.

None.
▼ Conclusion

The fitting was produced from Mn-containing Thomas steel. It was forged out of four strips, hot rolled and annealed. The corrosion contains only few external markers such as sand grains and dust particles in the outermost layers. The presence of soil materials in the welding seams near the surface could be due either to the corrosion progress (by diffusion through the corrosion crust) or to the manufacturing process.

#### ▼ References

#### References on object and sample

#### References object

1. Degrigny, C. (2011) Protection temporaire d'Objets métalliques base fer et cuivre à l'aide d'Inhibiteurs de corrosion Non Toxiques : application aux objets patrimoniaux techniques et scientifiques de grandes dimensions exposés en atmosphère non contrôlée, rapport interne HE Arc CR.

#### References sample

2. Rumo, L. (2009) Rapport Empa.

#### References on analytic methods and interpretation

ASTM E112-13: Standard Test Methods for Determining Average Grain Size.