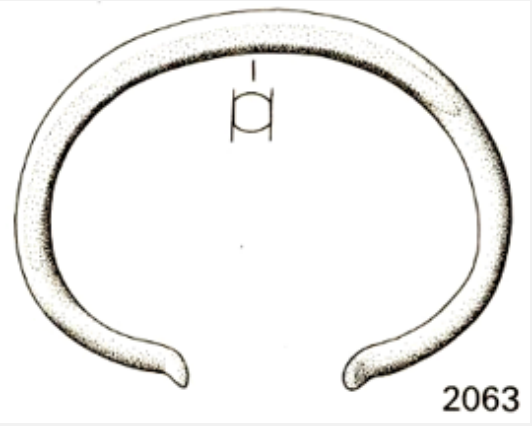
OVAL BRACELET WITH A ROUNDED DIAMETER B B3474 - LEADED BRONZE - LATE BRONZE AGE - SWITZERLAND

Artefact name Oval bracelet with a rounded diameter B B3474

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Url /artefacts/974/

The object



Credit HE-Arc CR.

Fig. 1: Oval bracelet with a rounded diameter (after Paszthory 1985, Tafel 171),

Description and visual observation

Description of the artefact Bracelet with a rounded diameter after Paszthory (1985, 243). Ends shaped to form little paws, with a casting seam on the inside (Fig. 1). Dimensions: Øobject = around 6.2cm.

Type of artefact Jewellery

Origin Les Eaux-Vives, Genève, Geneva, Switzerland

Recovering date Unknown

Chronology category Late Bronze Age

chronology tpq 1000 B.C.

chronology taq ----

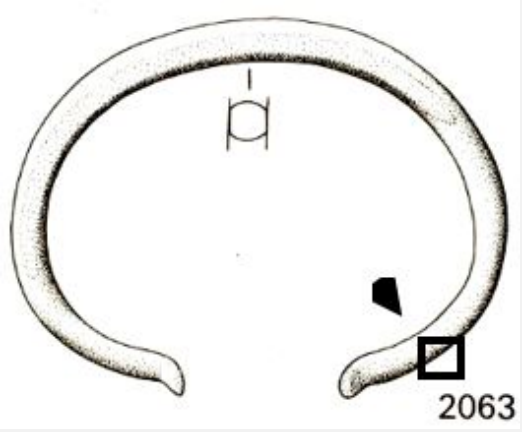
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Chronology comment	Hallstatt B2/3 (1000BC _ not defined)
Burial conditions / environment	Lake
Artefact location	Musées d'art et d'histoire, Genève, Geneva
Owner	Musées d'art et d'histoire, Genève, Geneva
Inv. number	B B3474
Recorded conservation data	N/A

Complementary information

None.

Study area(s)



Credit HE-Arc CR.

Fig. 2: Location of sampling area,

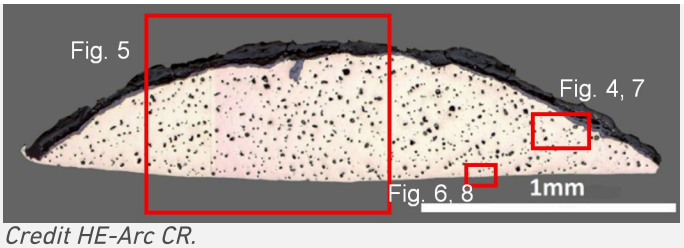
Binocular observation and representation of the corrosion structure

None.

MiCorr stratigraphy(ies) – Bi

Sample(s)

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Fig. 3: Micrograph of the cross-section of the sample taken from the bracelet with round diameter showing the location of Figs. 4 to 8,

Description of sample	The sample is a section from one end of the bracelet (Fig. 2). Its dimensions are: L = 2.3mm and W = 0.55mm. The corrosion layer is relatively thin (Fig. 3).
Alloy	Leaded Bronze
Technology	As-cast
Lab number of sample	MAH 77-110-2
Sample location	Musées d'art et d'histoire, Genève, Geneva
Responsible institution	Musées d'art et d'histoire, Genève, Geneva
Date and aim of sampling	1977 and 1991, study of the corrosion layer, metal composition

Complementary information

None.

Analyses and results

Analyses performed:
Metallography (etched with ferric chloride reagent), Vickers hardness testing, ICP-OES, SEM/EDS.

Non invasive analysis

None.

Metal

The remaining metal is a porous leaded bronze (Table 1). Under bright field light Pb and dark-grey copper sulphide inclusions can be seen (Fig. 4, Table 2). The copper sulphide inclusions are rather small with various forms, while the Pb inclusions are generally larger and round. The etched leaded bronze has the dendritic structure of an as-cast metal (Fig. 5) with an average hardness of HV1 80. After etching the Pb-inclusions turned dark grey and the copper sulphide light grey (Fig. 5).

Elements	Cu	Sn	Pb	Sb	As	Ni	Ag	Co	Zn	Fe	Bi

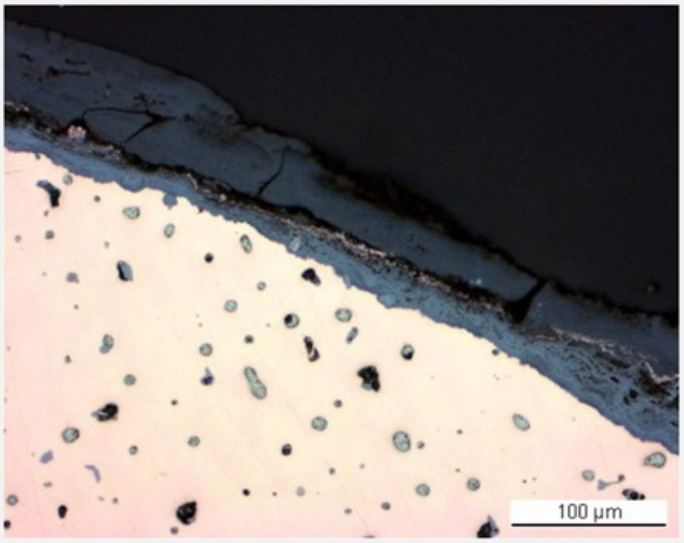
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mass%	89.67	6.40	2.62	0.52	0.27	0.22	0.13	0.07	0.03	0.04	0.03

Table 1: Chemical composition of the metal. Method of analysis: ICP-OES, Laboratory of Analytical Chemistry, Empa.

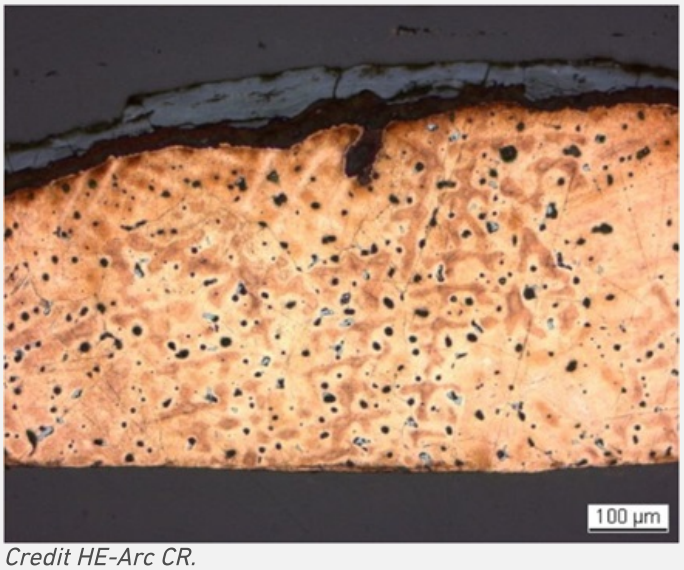
Elements	S	Cu	Total
Dark-grey inclusion	21	76	97

Table 2: Chemical composition (mass %) of dark-grey inclusions on Fig. 4. Method of analysis: SEM/EDS, Laboratory of Analytical Chemistry, Empa.



Credit HE-Arc CR.

Fig. 4: Micrograph of the metal sample from Fig. 3 (detail), unetched, bright field. In pink the metal with porosity (black), light-grey lead inclusions and dark-grey copper sulphide inclusions,



Credit HE-Arc CR.

Fig. 5: Micrograph of the metal sample from Fig. 3 (detail), etched, bright field. The dendritic structure is revealed with dark-grey lead inclusions and porosities (in black),

Microstructure	Dentritic structure with pores and inclusions
First metal element	Cu
Other metal elements	Ni, As, Ag, Sn, Sb, Pb

Complementary information

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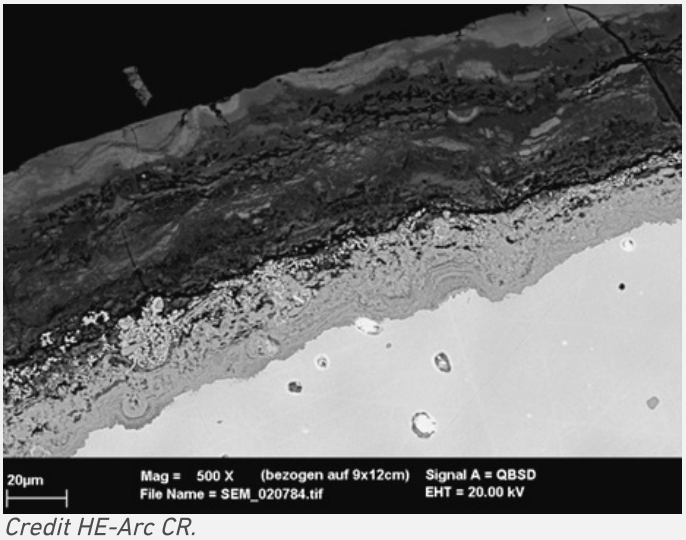
None.

Corrosion layers

The corrosion crust has an average thickness of 60µm and is composed of two main layers (CP2 and CP3, Fig. 4). In bright field, the inner layer (CP3) has a slight blue hue (Fig. 4). The corrosion products are stratified and the outer part is porous (Fig. 6). In polarised light, the inner layer appears as a mixture of reddish and orange corrosion products (Fig. 8). This layer is Cu- and O-rich with some Sn, Fe and Si in the porous zone (Table 3, Fig. 8). In bright field, the outer cracked and stratified layer (CP2) is dark grey (Fig. 5). It is depleted of Cu and rich in Fe and Sn with significant amounts of O and Si (Fig. 8). In polarised light, large red angular crystals (possibly cuprite) appear clearly in this outer corrosion layer (Fig. 7). The surface of the outermost layer (CP1) is Sn and O and Fe enriched (CP1, Fig. 8).

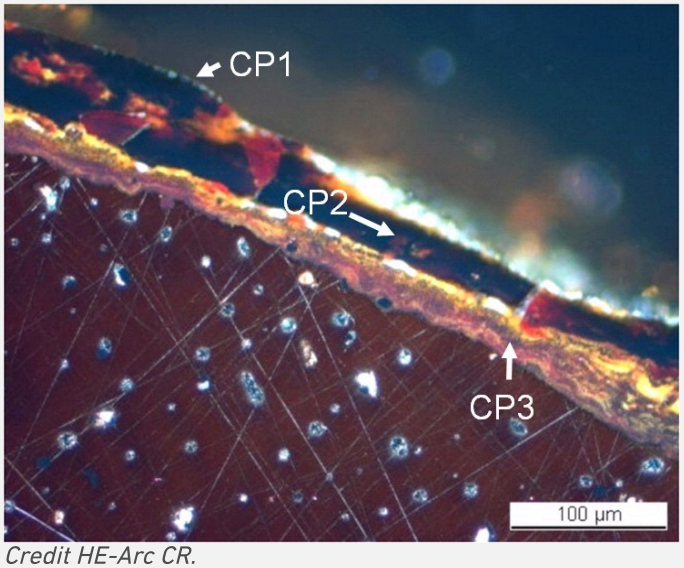
Elements	O	Fe	Cu	Sn	Pb	Si	S	As	Total
CP2	40	30	<	25	3.1	5	<	0.6	104
CP3	33	42	3.7	8.8	2.6	4.4	<	0.6	95
Red angular crystals in CP2	17	4.3	69	9.9	0.8	1.5	<	<	103

Table 3: Chemical composition (mass %) of the corrosion layers from Figs. 7 and 8. Method of analysis: SEM/EDS, Laboratory of Analytical Chemistry, Empa.



Credit HE-Arc CR.

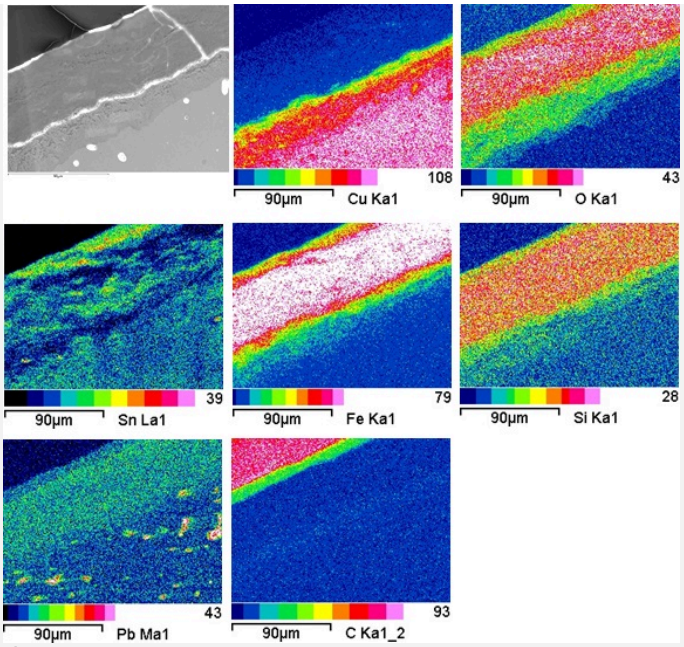
Fig. 6: SEM picture (detail of Fig. 3), BSE-mode. From bottom right to top left: the metal, the inner light-grey layer, the outer dark-grey layer with a lighter top zone,



Credit HE-Arc CR.

Fig. 7: Micrograph similar to Fig. 4 and corresponding to the stratigraphy of Fig. 9, polarised light. From bottom left to top right: the metal (in brown with white porosities and blue Pb inclusions), the inner layer (CP3) appearing as stratified red and orange layers followed by an outer layer (CP2, black with red angular crystals) and a superimposed light green layer (CP1),

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Credit Empa.

Fig. 8: SEM image, SE-mode, and elemental chemical distribution of a selected area of Fig. 6. The Pb mapping includes the Pb and copper sulphide inclusions, because of a peak interference. Method of examination: SEM/EDS, Laboratory of Analytical Chemistry, Empa,

Corrosion form	Uniform - pitting
Corrosion type	Type I (Robbiola)

Complementary information

None.

MiCorr stratigraphy(ies) – CS

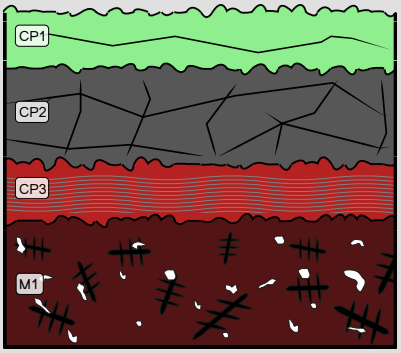


Fig. 9: Stratigraphic representation of the sample taken from the bracelet with round diameter in cross-section (dark field) using the MiCorr application. The characteristics of the strata are only accessible by clicking on the drawing that redirects you to the search tool by stratigraphy representation. This representation can be compared to Fig. 7, Credit HE-Arc CR.

Synthesis of the binocular / cross-section examination of the corrosion structure

None.

Conclusion

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The leaded bronze artefact shows an as-cast structure. The outer layer is typical for a lake patina (though in this case formed under aerobic conditions), containing principally Fe as well as other contextual elements (such as Si). The absence of Cu in the corrosion layer could be due to its re-crystallisation in large cuprite crystals. Surprisingly the top of the outer layer is enriched in Sn, which was not the case in the study carried out by Schweizer (Schweizer 1994). The additional presence of C in this top layer could indicate a secondary, terrestrial patina formation phase. The corrosion is a type 1 according to Robbiola et al. 1998.

References

References on object and sample

Reference object

1. Paszthory, K. (1985) Der bronzezeitliche Arm- und Beinschmuck in der Schweiz. PrähistorischeBronzefunde X-Bd. 3, München, 243, Tafel 171.

Reference sample

2. Empa report 137'695/1991, P. Boll.
3. Rapport d'examen (1977 and 1991) Laboratoire Musées d'art et d'histoire, Genève (1977-110).

References on analytic methods and interpretation

4. Robbiola, L., Blengino, J-M., Fiaud, C. (1998) Morphology and mechanisms of formation of natural patinas on archaeological Cu-Sn alloys, Corrosion Science, 40, 12, 2083-2111.
5. Schweizer, F. (1994) Objets en bronze provenant de sites lacustre: de leur patine à leur biographie. In: L'œuvre d'art sous le regard des sciences (éd. Rinuy, A. and Schweizer, F.), 143-157.

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