AXE HEAD N11 - TIN BRONZE - BRONZE AGE

Artefact name Axe head N11

Authors Christian. Degrigny (HE-Arc CR, Neuchâtel, Neuchâtel, Switzerland) & Valentin. Boissonnas (HE-Arc CR, Neuchâtel, Neuchâtel, Switzerland)

Url /artefacts/975/

The object



Credit HE-Arc CR, E.Forster.

Fig. 1: Axe head n°11,

Description and visual observation

Description of the artefact Axe head covered with a brown patina (Fig. 1). A uniform darker green corrosion product as well as a green powdery corrosion product appear locally in some areas. Dimensions: L = 90 mm; W max = 40 mm ; WT = 154.32 g.

Type of artefact Tool

Origin Unknown

Recovering date Unknown

Chronology category Bronze Age

chronology tpq ----

chronology taq ----

Chronology comment

This work is licensed under CC BY-NC-ND 4.0

Burial conditions / environment	Soil
Artefact location	Neues Museum, Biel/Bienne
Owner	Neues Museum, Biel/Bienne
Inv. number	N11
Recorded conservation data	The object might have been treated in the past.

Complementary information

None.

Study area(s)



Credit HE-Arc CR, E.Forster.

Fig. 2: General view of the axe head showing the location of sample 1 (Fig. 4),



Credit HE-Arc CR, E.Forster.

Fig 3: Side view of the axe head showing the location of sample 2 (Fig. 5), sample 3 (Fig. 6) and sample 4 (Fig. 7),



Fig 4: Local corrosion (diameter: 2 mm) where sample 1 has been taken,

This work is licensed under CC BY-NC-ND 4.0

Credit HE-Arc CR, E.Forster.



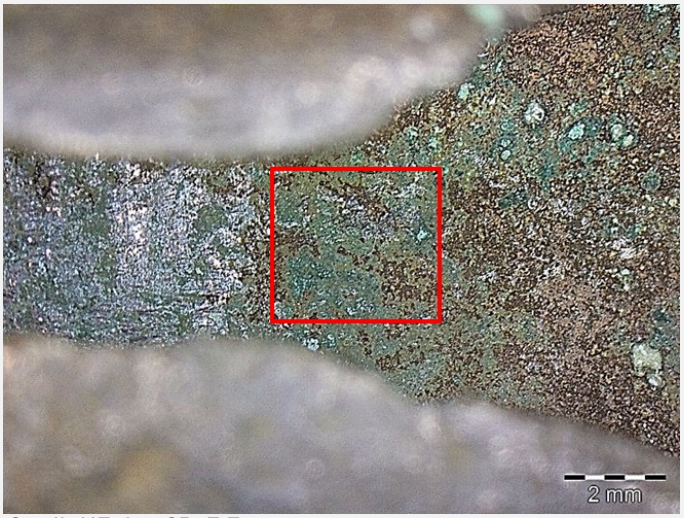
Credit HE-Arc CR, E.Forster.

Fig. 5: Local corrosion (length max: 30 mm) where sample 2 has been taken,



Credit HE-Arc CR, E.Forster.

Fig. 6: Local corrosion (length max: 10 mm) where sample 3 has been taken,



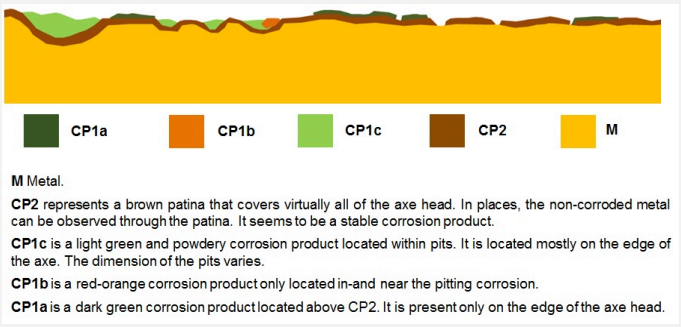
Credit HE-Arc CR, E.Forster.

Fig 7: Green dark corrosion layer where sample 4 has been taken,

Binocular observation and representation of the corrosion structure

The schematic representation below gives an overview of the corrosion layers encountered on the head axe from a first visual macroscopic observation.

This work is licensed under CC BY-NC-ND 4.0



Credit HE-Arc CR, E.Foster.

Fig. 8: Stratigraphic representation of corrosion products of the axe head (based on visual observations),

MiCorr stratigraphy(ies) – Bi

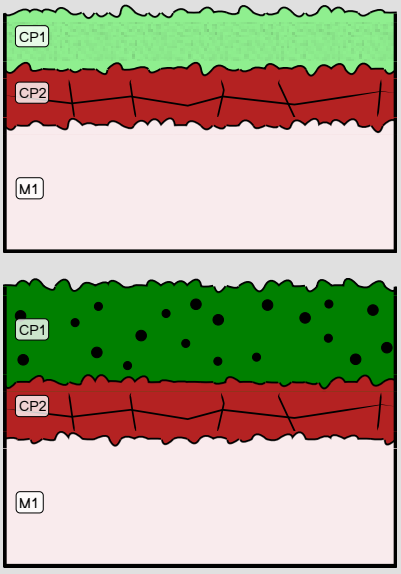
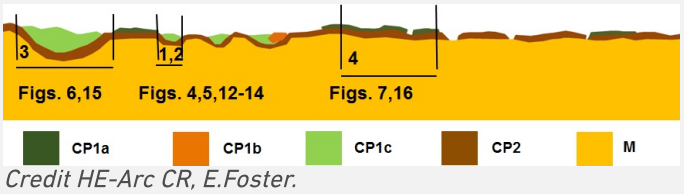


Fig. 9: Stratigraphic representation of the corrosion structure of the axe (sample 3) observed macroscopically under binocular microscope using the MiCorr application. The characteristics of the strata are only accessible by clicking on the drawing that redirects you to the search tool by stratigraphy representation. This presentation can be compared to Fig. 18. CP1 corresponds to CP1c, Credit HE-Arc CR, C.Degrigny.

Fig. 10: Stratigraphic representation. of the corrosion structure of the axe (sample 4) observed macroscopically under binocular microscope using the MiCorr application. The characteristics of the strata are only accessible by clicking on the drawing that redirects you to the search tool by stratigraphy representation. This presentation can be compared to Fig. 18. CP1 corresponds to CP1a, Credit HE-Arc CR, C.Degrigny.

Sample(s)



Credit HE-Arc CR, E.Foster.

Fig. 11: Location of the samples on the stratigraphic representation of Fig. 8 with indication of the corresponding figures,

Description of sample	Particles were sampled (Figs. 4-7) with a scalpel from the areas where local corrosion has developed.
Alloy	Tin Bronze
Technology	Unknown
Lab number of sample	-----

This work is licensed under CC BY-NC-ND 4.0

Sample location	HE-Arc CR, Neuchâtel, Neuchâtel
Responsible institution	HE-Arc CR, Neuchâtel, Neuchâtel
Date and aim of sampling	March 04, 2013, chemical and structural analysis

Complementary information

None.

Analyses and results

Analyses performed:
SEM-EDS, FTIR. The FTIR device used is a Biorad Excalibur FTS 3000 spectrometer coupled to an IR microscope UMA500. The measurements were performed between 4000 and 650 cm-1.

Non invasive analysis

None.

Metal

SEM/EDS analysis of corroded particles indicates that the alloy might be composed of copper (Cu) and tin (Sn) with some arsenic (As). The core metal is most likely a tin bronze.

Microstructure	Unknown
First metal element	Cu
Other metal elements	As, Sn

Complementary information

None.

Corrosion layers

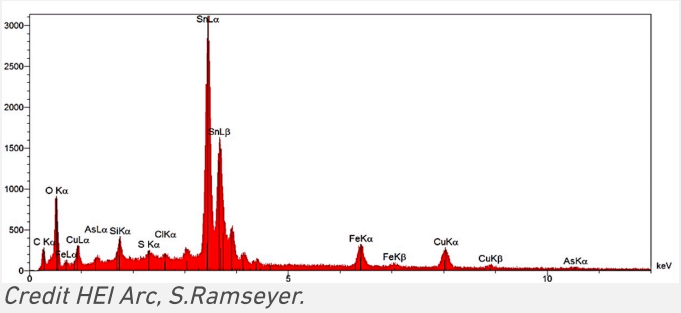
Analyses by SEM/EDS indicate that the light yellow particles of sample 1 are predominantly composed of Sn and O with minor presence of C, Cu, Fe, S, Cl and As (Fig. 12), while the green and red particles of sample 2 are respectively composed of O, Cu, Cl, and Sn, with a minor presence of Fe and S (Fig. 13) and Cu, O (Fig. 14, possibly cuprite). Sample 3 consists of green particles composed mostly of Sn, Cl, Cu and O with S, Al and Fe as minor elements (Fig. 15). Sample 4 is constituted of green and white particles. The former have the same composition as the particles of

This work is licensed under CC BY-NC-ND 4.0

sample 3 while the white particles are significantly different in composition: containing a large amount of C with a small amount of O, Cl, S, K and Cu (Fig. 16). The FTIR spectrum of the white particles (Fig. 17) shows several characteristic peaks of an organic compound: 2919 cm-1 and 2850 cm-1 (C-H- bond), 3445 cm-1 and 3323 cm-1 (O-H- bond) or 1030cm-1, 1006cm-1 and 951cm-1 (C-O-C- bond).

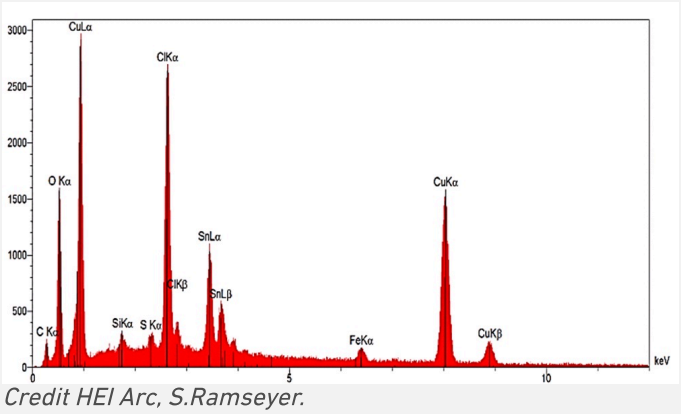
	Sample / particles	Major elements	Minor elements
Green powdery corrosion product	Sample 1	Sn + O	C + Cu + Fe + S + Cl + As
	Sample 2 – CP1c (Fig. 11) - Green particles	O + Cu + Cl + Sn	Fe + S
	Sample 2 – CP2 (Fig.11) - Red-brown particles	Cu + O	
	Sample 3 - CP1c (Fig. 11)	Sn + Cl+ Cu + O	S + Al + Fe
Darker green corrosion product	Sample 4 – CP1a (Fig. 11) - Green particles	Cl + Sn + Cu + O	S + C
	Sample 4 – CP1a (Fig. 11) - White particles	C	O + Cl + S + K + Cu

Table 1: Identification of chemical elements in the corrosion products on the head axe. Method of analysis: SEM/EDS, HEI-Arc.



Credit HEI Arc, S.Ramseyer.

Fig, 12: EDS spectrum of sample 1 (yellow powder),



Credit HEI Arc, S.Ramseyer.

Fig. 13: EDS spectrum of sample 2 (green particle),

This work is licensed under CC BY-NC-ND 4.0

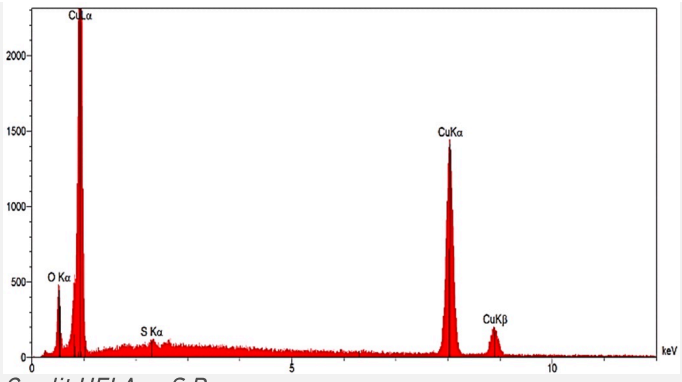


Fig. 14: EDS spectrum of sample 2 (red particle),

Credit HEI Arc, S.Ramseyer.

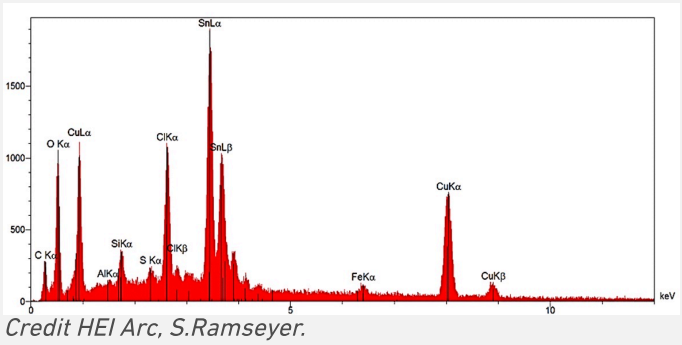


Fig. 15: EDS spectrum of sample 3 (green corrosion product),

Credit HEI Arc, S.Ramseyer.

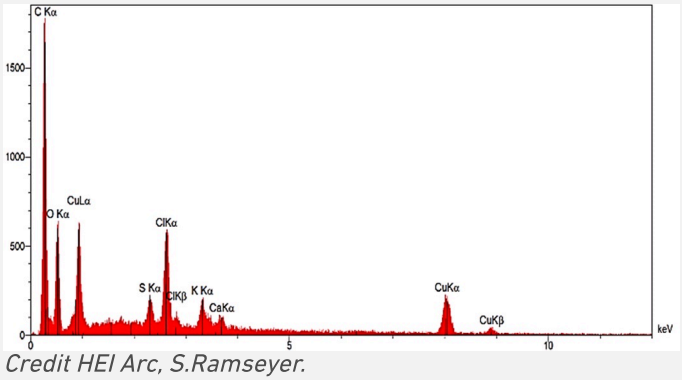


Fig. 16: EDS spectrum of sample 4 (white particle),

Credit HEI Arc, S.Ramseyer.

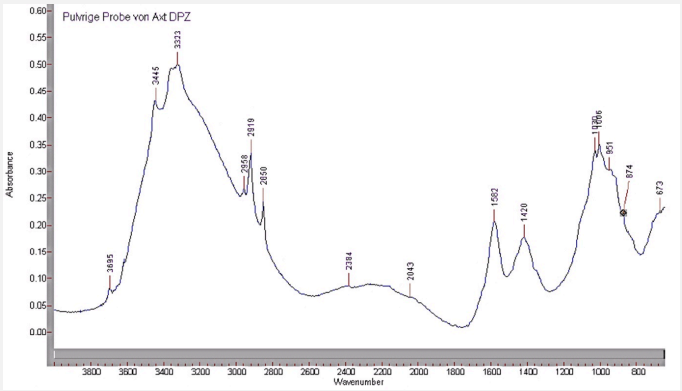


Fig. 17: FTIR spectrum of sample 4 (white particle),

Credit Lab of Swiss National Museum, Affoltern am Albis.

Corrosion form	Multiform - pitting
Corrosion type	Unknown

Complementary information

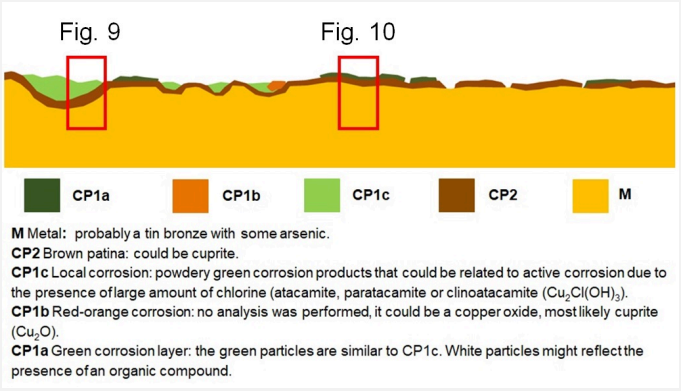
None.

This work is licensed under CC BY-NC-ND 4.0

MiCorr stratigraphy(ies) – CS

Synthesis of the binocular / cross-section examination of the corrosion structure

The schematic representation of corrosion layers of Fig. 8 integrating additional information based on the analyses carried out is given in Fig. 18.



Credit HE-Arc CR, E.Foster

Fig. 18: Improved stratigraphic representation of a virtual cross-section through the axe head based on visual observations and additional analyses and visualization of the two stratigraphies of Figs. 9 and 10.

Conclusion

The axe head is probably a tin bronze containing a small amount of As. It is difficult to interpret the corrosion layer only by macroscopic observation. The analyses show that all green corrosion products contain chlorides that might originate from strip-cleaning by hydrochloric acid (although not documented). Active corrosion is not proved but we suspect the presence of atacamite, paratacamite or clinoatacamite (Cu2Cl(OH)3). FTIR revealed the presence of an organic compound on the metal surface (possibly a protective coating).

References

References on analytic methods and interpretation

1. Bertholon, R. (2000) La limite de la surface d'origine des objets métalliques archéologiques. Caractérisation, localisation et approche des mécanismes de conservation. ©Régis Bertholon, Paris.
2. Degrigny, C. and Senn, M. (2012) Methodology to study and analyse the microstructures and corrosion forms of ancient and historic metals: application to metallographic samples from Swiss collections MIFAC-Métal. Projet Sagex, Final Report, Haute école de conservation-restauration ARC, Neuchâtel.
3. Selwyn, L. (2004) Métaux et corrosion Un manuel pour le professionnel de la conservation. Institut Canadien de Conservation, Ottawa.

This work is licensed under CC BY-NC-ND 4.0